



## Speeding up $^{17}\text{O}$ MQMAS Data Collection for D-glucose by Paramagnetic Doping

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### Introduction

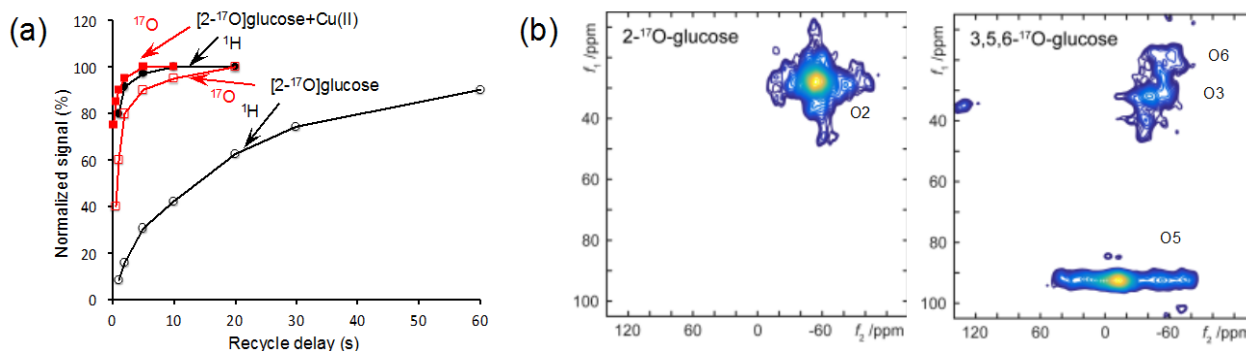
Half-integer nuclei such as  $^{17}\text{O}$  ( $I = 5/2$ ) usually have significantly shorter spin-lattice ( $T_1$ ) relaxation times than spin-1/2 nuclei, such as  $^1\text{H}$  and  $^{13}\text{C}$  because of the large nuclear quadrupole interactions. However, there are also well-documented cases where relaxation times for quadrupolar nuclei in organic solids can also be rather long. For example, the  $T_1(^{17}\text{O})$  value in potassium hydrogen [ $^{17}\text{O}_4$ ]maleate is about 10 s. It is well known that D-Glucose related compounds have long  $T_1(^1\text{H})$  values. Recently, we discovered that the  $T_1(^{17}\text{O})$  values in  $^{17}\text{O}$ -glucose compounds are also long. In [2- $^{17}\text{O}$ ]-D-glucose, for example,  $T_1(^1\text{H})$  and  $T_1(^{17}\text{O})$  values are ca. 30 and 5 s, respectively. Apart from synthetic difficulties in  $^{17}\text{O}$ -labeling, carbohydrate molecules are also the least studied by  $^{17}\text{O}$  solid-state NMR [1]. Paramagnetic doping has long been used to shorten the relaxation times for spin-1/2 nuclei such as  $^1\text{H}$  [2]. Here, we applied the same method to shorten  $T_1(^{17}\text{O})$  in D-glucose compounds so that 2D  $^{17}\text{O}$  3QMAS data collection can be significantly accelerated.

### Experimental

D-glucose compounds were co-dissolved with  $\text{Na}_2\text{Cu}(\text{II})\text{EDTA}$  (10% wt) in aqueous solution. Greenish solids were obtained by drying the solution with a stream of  $\text{N}_2$ .  $^{17}\text{O}$  solid-state NMR spectra were acquired at 19.6 T at the NHMFL, using a Bruker NEO console and a 3.2 mm Low-E MAS probe designed and built at the NHMFL.

### Results and Discussion

Figure 1(a) shows that, upon paramagnetic doping, the  $^1\text{H}$  and  $^{17}\text{O}$  relaxation times of [2- $^{17}\text{O}$ ]-D-glucose were significantly shortened. In fact, both  $T_1(^1\text{H})$  and  $T_1(^{17}\text{O})$  values are now less than 1 s. The  $^{13}\text{C}$  CPMAS spectra of the doped glucose compounds do not display any noticeable difference from those of undoped compounds. Now the  $^{17}\text{O}$  NMR data can be accumulated at a much faster rate (e.g., with a recycle delay as short as 0.5 s). With this improvement, we were able to acquire 2D  $^{17}\text{O}$  3QMAS spectra for [2- $^{17}\text{O}$ ]-D-glucose and [3,5,6- $^{17}\text{O}_3$ ]-D-glucose, as shown in Figure 1(b). Interestingly, we note immediately that the hydroxyl groups (O2, O3 and O6) exhibit much shorter  $T_2$  values (thus poorer resolution in the isotropic dimension) than the non-protonated oxygen (O5). However, given that  $^1\text{H}$  decoupling of >60 kHz was employed during data acquisition, it is surprising to see such a  $T_2$  discrepancy among different sites. It is possible that it is due to the high-order decoupling-induced recoupling effect [3] or unknown dynamics in doped D-glucose. We are currently investigating this phenomenon.



**Fig. 1** (a) Effects of paramagnetic doping with  $\text{Na}_2\text{Cu}(\text{II})\text{EDTA}$  on the relaxation times of [2- $^{17}\text{O}$ ]-D-glucose. Data were obtained at 14.1 T. (b)  $^{17}\text{O}$  shifted-echo 3QMAS spectra of [2- $^{17}\text{O}$ ]-D-glucose and [3,5,6- $^{17}\text{O}_3$ ]-D-glucose obtained at 19.6 T. The sample spinning frequency was 16 kHz. In each spectrum, a total of 16 complex  $t_1$  points were collected. The numbers of transients were 13632 and 62400 for [2- $^{17}\text{O}$ ]-D-glucose and [3,5,6- $^{17}\text{O}_3$ ]-D-glucose, respectively.

### Acknowledgements

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### References

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- [3] Wu, G., Chem. Phys. Lett., **322**, 513 (2000).