



Unifying Concepts Linking Dissolved Organic Matter Composition to Persistence in Aquatic Ecosystems

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Summary

The link between composition and reactivity of dissolved organic matter (DOM) is central to understanding the role aquatic systems play in the global carbon cycle; yet, unifying concepts driving molecular DOM composition across aquatic ecosystems have yet to be established. We characterized 37 DOM isolates from diverse aquatic ecosystems, including their stable and radiocarbon isotopes ($\delta^{13}\text{C}$ - dissolved organic carbon (DOC) and $\Delta^{14}\text{C}$ -DOC), optical properties (absorbance and fluorescence), and molecular composition (using the custom-built 9.4 tesla Fourier transform ion cyclotron resonance mass spectrometer with negative electrospray ionization at the National High Magnetic Field Laboratory in Tallahassee, Florida). Isolates encompassed end-members of allochthonous and autochthonous DOM from sites across the United States, the Pacific Ocean, and Antarctic lakes. Modern $\Delta^{14}\text{C}$ -DOC and optical properties reflecting increased aromaticity, such as carbon specific UV absorbance at 254 nm (SUVA₂₅₄), were directly related to polyphenolic and polycyclic aromatic compounds, whereas enriched $\delta^{13}\text{C}$ -DOC and optical properties reflecting autochthonous end-members were positively correlated to more aliphatic compounds. Furthermore, the two sets of autochthonous end-members (Pacific Ocean and Antarctic lakes) exhibited distinct molecular composition due to differences in extent of degradation. Across all sites and end-members studied, we find a consistent shift in composition with aging, highlighting the persistence of certain biomolecules concurrent with degradation time.

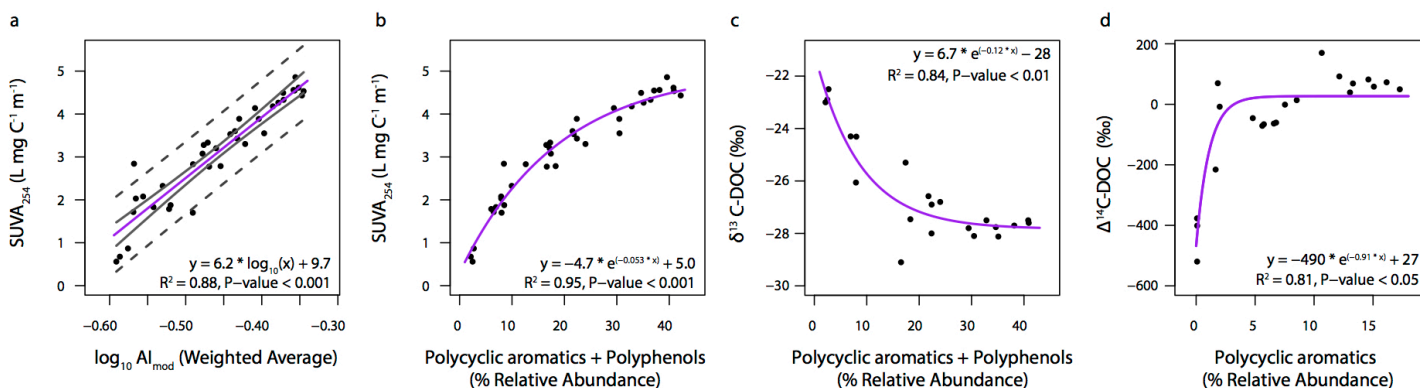


Fig. 1: Relationships between molecular composition, optical properties and isotopic content. Changes in SUVA₂₅₄ against (a) log₁₀ transformed modified aromaticity index and (b) percent relative abundance of polycyclic and polyphenolic aromatic compounds (n = 37), and changes in isotopic composition against molecular composition; (c) $\delta^{13}\text{C}$ -DOC against percent relative abundance of polycyclic and polyphenolic aromatic compounds and (d) $\Delta^{14}\text{C}$ -DOC against polycyclic aromatic compounds (n = 21).

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References

- [1] Kellerman, A.M., *et al.*, Environmental Science & Technology, **52**, 2538-2548 (2018).