

35T SCH NMR ¹⁷O Spectroscopy Reveals Long-Term Water Stability in a Transmembrane Pore

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Introduction

Gramicidin A (gA) is dimeric ion channel with a 20 Å long pore supporting a single water file across liquid crystalline bilayers [1,2]. We have performed ¹⁷O NMR of gA using the SCH Magnet at 35.2T and report here that the electric dipole orientation of the water wire within the pore is orders of magnitude more stable than had been predicted by molecular dynamics (MD) simulations [2].

Experimental

gA ¹⁷O single site labeled at Gly₂, Leu₄, Leu₁₀, Leu₁₂ or Leu₁₄ in DMPC bilayers were aligned as described previously [3ab]. Oriented sample solid-state NMR (OSssNMR) spectra were acquired at the SCH magnet at the NHMFL in Tallahassee. The SCH was set at 35.2T, 203 MHz ¹⁷O field. An NMR probe optimized for static samples developed for the SCH magnet by the RF group in the NMR facilities of the NHMFL was used. The spectra was acquired with a DFS-echo pulse sequence with a 90° solid pulse of 1.5 µs and 20 µs echo time in the presence of 30 kHz ¹H decoupling.

Results and Discussion

The OSssNMR spectra of gA ¹⁷O Gly₂, Leu₄ and Leu₁₀ show two well-resolved signals (Fig. 1). The same was not true for Leu₁₂ and Leu₁₄ where a single narrow resonance was observed for each site. While Gly₂, Leu₄ and Leu₁₀ are exposed to the single file water wire, Leu₁₂ and Leu₁₄ carbonyls are fully exposed to the aqueous environment. Density functional theory (DFT) simulations in good agreement, reproduce the experimental peak distances observed in the NMR spectra (Table I). This confirms that the peak doubling is due to asymmetrical hydrogen bonding between residues in each monomer and the water wire, which is stable on the millisecond time scale. MD simulations that show the waters in the channel flipping at the subnanosecond time scale cannot correctly model this result.

Conclusions

The asymmetric distribution of water hydrogen bonding to the pore lining carbonyls of the two monomers of the gA backbone causes doubling of the resonances, breaking the antiparallel dimeric symmetry of gA. This is confirmed by DFT calculations. The stability of the water wire orientation is on the millisecond timescale in that, if the water wire flipped on the sub msec timescale, there would be averaging of the resonance pairs and only a single peak would be observed in the spectra. This is the first time that longterm water stability has been reported within a channel.

Acknowledgements

This work was funded in part by NIH P41 GM122698. The1500 MHz NMR console was funded by NSF DMR-1039938. This work was performed at the

NHMFL, which is supported by the National Science Foundation through NSF/DMR-1157490/1644779 and the State of Florida.

References

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Figure 1. Gly₂, Leu₄, Leu₁₀, Leu₁₂ and Leu₁₄ carbonyl oxygens were individually labeled with ¹⁷O such that each sample was labeled in both gA monomers with a single isotopic label. Peaks corresponding to residues where a stable hydrogen bond with water take place (DFT assignment) are marked with a *.

Table I. Distance (in ppm) between peaks inNMR spectra and resonances calculated fromDFT simulation.

Residue	NMR	DFT
Gly ₂	23.2 ± 2.5	25.9
Leu ₄	6.9 ± 2	4.8
Leu ₁₀	14 ± 2.8	8.7