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Speciation of Metals in Asphaltenes by High-Performance Thin-Layer Chromatography and Solid—Liquid Extraction Hyphenated with Elemental and Molecular Identification

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ABSTRACT: Asphaltenes are among the most challenging components in petroleum processing because they contain high amounts of heteroatoms (i.e., S, N, O, V, and Ni) thought to be responsible for strong aggregation tendencies, precipitation, and fouling problems. The role of vanadium- and nickel-containing petroleum compounds (i.e., petroporphyrins) in aggregation and fouling is not completely understood because asphaltene composition and structure is still a subject of debate in the petroleum chemistry community. Characterization of asphaltenes, namely, molecular analysis that employs no chromatographic separation, often fails to reveal their comprehensive composition. The work herein presents asphaltene fractionation by (1) solid/liquid extraction, which allows for separation of single-core ("island") and multicore ("archipelago") structural motifs and by (2) highperformance thin layer chromatography (HPTLC) with cellulose as the stationary phase and DCM/MeOH as the eluent, which facilitates access to petroporphyrins. Characterization is performed by laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) and matrix-assisted laser desorption ionization Fourier transform ion cyclotron resonance mass spectrometry (MALDI FT-ICR MS). The results demonstrate that even with multiple separation steps, a large quantity of vanadyl porphyrins remains inaccessible for molecular analysis by MALDI FT-ICR MS, which raises the question of what portion of a complex sample of asphaltene can be revealed by ultrahigh resolution mass spectrometry. Furthermore, the results show that easily accessible porphyrins migrate with the solvent front in HPTLC. Thus, HPTLC can be used to isolate and identify "free" porphyrins not locked into asphaltene aggregates; however, further development of separation methods is required to access the most difficult and problematic asphaltene fractions, which do not migrate and impose analytical challenges due to their stronger aggregation tendency.

■ INTRODUCTION

Asphaltenes are well known for their negative impact on petroleum production, transportation, storage, and refining. Asphaltenes contain a wide diversity of molecules with different degrees of aromaticity, heteroatom contents, and molecular structure. For instance, Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR MS) has revealed asphaltene molecules with either single-core or multicore structures that contain only hydrogen and carbon (hydrocarbons or the HC class) as well as species with up to three or more heteroatoms (e.g., vanadyl porphyrins or the $N_4O_1V_1$ class, sulfur-containing molecules or the S_3 class, and oxygen-rich compounds or the O_4 class).¹⁻⁶ Thus, the combined effects of aromaticity, heteroatom content, and molecular structure trigger multiple intermolecular interactions, such as pi-stacking, hydrogen bonding, acid-base conjugations, and London dispersion forces between asphaltenes. The synergy between multiple intermolecular interactions results in strong aggregation and ultimate precipitation of a wide compositional range of species, from those with no heteroatoms (HC class) to high heteroatom-containing species.⁷⁻⁹ It is believed that metals present in asphaltenes are mainly present in porphyrinic complexes,¹⁰⁻¹² i.e., nickel

and vanadyl porphyrins are tetrapyrrole-based metal complexes. The actual structures of nickel and vanadyl porphyrins are the result of geochemical modifications of the chlorophyll molecule. The chlorophyll structure changed over geological time during fossil crude oil formation via demetalation, aromatization, and peripheral addition of sulfur or oxygen.^{13,14} These transformations yielded several structurally different porphyrin families of the types $N_4O_1V_1$, $N_4O_2V_1$, $N_4O_3V_1$, $N_4O_1V_1S_1$, and N_4Ni_1 , as shown in Figure 1.

Understanding the molecular composition of porphyrins and other metal complexes that trigger issues in refinery processes is critical to the petroleum industry. For this purpose, it is essential to fractionate asphaltenes before molecular-level analysis because direct "dilute-and-shoot" studies of such a complex matrix usually do not reveal a great part of the information. As shown by Evdokimov et al., ^{15,16} even with the

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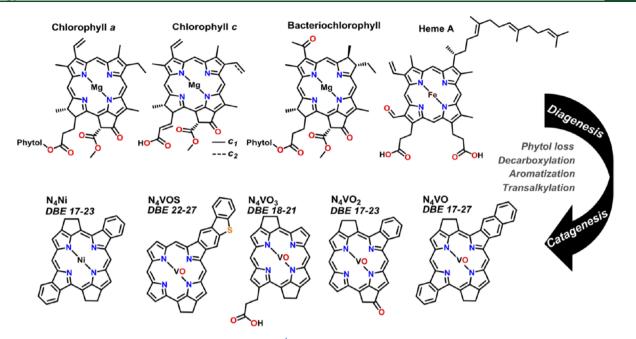


Figure 1. Scheme of geochemical transformation of chlorophylls.⁴

abundant presence of petroporphyrins, the Soret band, characteristic of porphyrinic complexes in ultraviolet absorption spectra, is not always detected in the sample spectrum, the detection of which depends on the dilution factor. Furthermore, with MS techniques such as matrix-assisted laser desorption ionization (MALDI) FT-ICR MS, the matrix, ionization efficiency, and ion suppression effects usually conceal the molecular level of difficult-to-ionize compounds. Recently, Chacón-Patiño et al.¹⁷ and Giraldo-Davila et al.¹⁸ used silica gel thin-layer chromatography plates to separate asphaltene samples and achieve a more complete molecular characterization. The authors used a mixture of DCM/MeOH, followed by toluene, to separate asphaltenes into three fractions enriched in alkyl-aromatic compounds, polar species, and non-eluted asphaltenes. This technique can also be used on the preparative scale to separate fractions that can then be extracted with solvents and further analyzed by other techniques (Li et al.¹⁹). Furthermore, asphaltene samples can also be separated prior to high-performance thin layer chromatography (HPTLC) analysis. For example, Chacón-Patiño et al.^{5,20,21} reported asphaltene fractionation by a solidliquid extraction strategy using silica gel as an adsorbent and subsequent Soxhlet extraction; this method is known as extrography. The authors demonstrated that extrography allows for the separation of asphaltenes according to their structural motif, multicore or archipelago (composed of several aromatic cores linked by covalent bonds) and single-core or island (composed of one aromatic core with alkyl side chains). According to these results, island compounds are easily accessed by atmospheric pressure photoionization FT-ICR MS; conversely, archipelago molecules are more difficult to ionize, due in part to their relatively stronger aggregation trends.²²

The use of TLC or HPLTC allows for more comprehensive elemental information for petroleum fractions.^{23,24} For instance, Chirinos et al.²⁵ used laser ablation ICP-MS to achieve rapid determination of the ratio between nickel and vanadium in asphaltene samples after a separation step with HPTLC. Moreover, the use of HPTLC plates also allows for

molecular analysis by LDI/MALDI FT-ICR MS; for example, Cho et al.²⁶ demonstrated that LDI FT-ICR MS can be used to quantify porphyrins in petroleum samples. Additionally, Pradilla et al.²⁷ developed a MALDI matrix to promote selective ionization or petroporphyrins via electron transfer reactions. The authors used a novel matrix to analyze asphaltenes after an HPTLC separation step. The results revealed the selective ionization of petroporphyrins and an increased number of molecular formulas compared to those identified in previous studies.⁴

In this work, two successive separation steps, extrography and HPTLC, were coupled with elemental laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) and molecular detection (MALDI FT-ICR MS) to characterize metals contained in one asphaltene reference sample, referred to as Asphaltene 2017. The purpose was to determine which fractions contained vanadium and the kinds of porphyrins present in these fractions.

EXPERIMENTAL SECTION

Samples and Reagents. *Samples.* Asphaltene 2017 (from a Middle Eastern crude oil; vanadium content: 640 ppm) from the interlaboratory study of the Petrophase 2017 Conference²⁸ and its extrography fractions, prepared as reported by Chacon-Patiño,^{5,20} were studied. Asphaltene 2017 was obtained by a slightly modified version of the ASTM D6560-12 method. A cleaning step, via four cycles of maceration and Soxhlet extraction with heptane, was employed to decrease the amount of occluded/coprecipitated maltenes.

Solvents. Tetrahydrofuran (THF, HPLC grade, without stabilizer, Lichrosolv, Merck, Frankfurter Strasse 250, Darmstadt 64293, Germany) was used as the solvent for sample preparation and for MALDI analyses. Dichloromethane (DCM) and methanol (MeOH) (HPLC grade, Lichrosolv, Merck, Frankfurter Strasse 250, Darmstadt 64293, Germany) were used to develop the HPTLC plates. Toluene, heptane, and pentane were used to prepare Petrophase 2017 asphaltenes. Acetone, toluene, heptane, THF, and methanol of HPLC grade were used to prepare extrography fractions from Asphaltene 2017.

Solid/Liquid Extraction Samples. Extrography fractionation was carried out following a method reported elsewhere.²⁰ Briefly, a Soxhlet

apparatus was used to perform three successive extractions. Approximately 150 mg of asphaltenes was dissolved in dichloromethane, and 15 g of silica was then added to the solution. Subsequently, the mixture was completely dried under nitrogen, and the dried mixture was Soxhlet extracted with acetone, toluene/ heptane (50/50 Hep/Tol, v/v), and toluene/tetrahydrofuran/ methanol (47.5/47.5/5 Tol/THF/MeOH, v/v/v).

Instrumentation. *HPTLC Plates.* Cellulose HPTLC glass plates (Merck, thickness 0.15-0.2 mm, Frankfurter Strasse 250, Darmstadt 64293, Germany) that were 20 cm \times 10 cm in size and did not contain fluorescent indicators were used. The HPTLC procedure was carried out with the following instrumentation protocol from CAMAG. Sample application and sample development were performed using an automatic TLC Sampler 4 (ATS4) and automated multiple development 2 (AMD2) equipment. UV absorbance chromatograms were acquired at a wavelength of 280 nm with a D2 and W lamp TLC scanner. UV spectra from 190 to 900 nm were obtained using the same device. Finally, a TLC visualizer was used to obtain images of the cellulose plate under 366 nm UV light after development. Peak area data were determined using WinCATS software.

HPTLC plates were cleaned with DCM and THF and dried at 110 $^{\circ}$ C for 10 min prior to sample deposition. All the samples were deposited 10 mm from the bottom of the plate. A high eluent-strength mobile phase (DCM/MeOH, 99.5:0.5) was used. The plate was developed in a single step until the solvent front reached 50 mm. Whole Petrophase 2017 and its extrography fractions were deposited two times per HPTLC plate on three different plates that were developed as described above for subsequent analysis by LA-ICP-MS.

Developed HPTLC separations for whole Petrophase 2017 and its three extrography fractions were also subjected to MALDI FT-ICR MS analysis. After separation, the different samples were scratched from the plates, and extraction using THF was performed and mixed with a custom synthesized MALDI matrix (matrix/sample ratio of 1) developed by the Combariza Research Group.^{4,27,29} The MALDI matrix facilitated porphyrin ionization by electron transfer reactions and is known as α -cyanophenylenevinylene (α -CNPV-CH3).

LA-ICP-MS. Element-specific analyses of the developed HPTLC plates were carried out by LA-ICP-MS using a New Wave Research UP-213 Nd:YAG laser (ESI, Fremont, CA) coupled with ICP-MS (7700 series, Agilent, Santa Clara, CA)²⁴ under the operating conditions given in Table 1. The instrument was equipped with a

Table 1. LA	A-ICP MS	Operating	Parameters
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Agilent 7700 Series ICP MS				
plasma Ar gas flow rate (L min ⁻¹)	15			
auxiliary Ar gas flow rate (L min ⁻¹)	1			
measured isotope	¹³ C, ³² S, ⁵¹ V, ⁵⁸ Ni, ⁶⁰ Ni			
analysis time (s)	506			
New Wave Research UP-213 LA				
wavelength (nm)	213			
fluence (J cm ⁻²)	0.85			
spot size (µm)	200			
scan speed (μ ms ⁻¹)	100			
carrier He gas flow rate $(L min^{-1})$	0.8			

Fassel-type quartz torch shielded with a grounded Pt electrode and a quartz bonnet; a standard quartz injector (1.75 mm i.d.) was used. An ablation chamber was installed on a three-axis translation stage and coupled to the ICP torch using a 60 cm Tygon tube (5.0 mm i.d.). The ablated material was swept by the carrier gas (helium) and mixed with the spectrometer makeup gas (argon) prior to introduction into the plasma. Ni sampler (orifice diameter: 1.1 mm) and skimmer (orifice diameter: 0.8 mm) cones were optimized for the analyses.²⁴ The laser beam followed a linear trajectory throughout the migration. To fit within the ablation cell, plates were cut to dimensions of 9 × 10

 cm^2 . Each sample on the three different plates was analyzed in triplicate.

MALDI FT-ICR MS. A hybrid quadrupole FT-ICR instrument (SolariXR, Bruker Daltonics, Bremen, Germany) equipped with a 12 T superconducting magnet was equipped with an LDI/MALDI source with a Nd:YAG × 3 solid-state laser (355 nm).³⁰ Broadband mass spectra were acquired over a mass range of m/z 147–1000 with an accumulation of 200 scans. The signal was digitalized with 8 M points, resulting in a transient length of 3.4 s; this gave a resolving power number of 0.9×10^6 at m/z 400. Instrument control and data acquisition were facilitated by DataAnalysis (ver. 4.4); OriginPro (ver. 2016) was used to process and visualize the data. From the molecular formulas determined from the accurate mass measurements (typically <0.2 ppm), double bond equivalent (DBE) values were calculated as shown in eq 1 (*c*, carbon number; *h*, hydrogen number; *n*, nitrogen number).³¹

$$DBE = c - \frac{h}{2} + \frac{n}{2} + 1$$
(1)

RESULTS

HPTLC Results. Four asphaltene samples, the whole Petrophase 2017 asphaltene sample (A2017) and its three extrography fractions (acetone, Hep/Tol, and Tol/THF/ MeOH), were separated via HPTLC and analyzed with a UV densitometer. UV analysis at 280 nm of the developed HPTLC plates (Figure 2 A) demonstrates that the samples were separated into two major fractions, eluted and non-eluted. Figure 2 B features the complete UV spectra for the eluted and non-eluted fractions. The results demonstrate that all noneluted fractions have a similar spectrum, in which the absorbance reaches a maximum at ~240 nm and then decreases continuously until ~800 nm. However, the eluted HPTLC fractions present different trends. For instance, the Tol/THF/MeOH extrography fraction features an absorption spectrum with a weaker signal between 400 and 600 nm than those of the other fractions. Eluted species from whole PetroPhase 2017 and its Hep/Tol extrography fraction present similar absorption spectra. Finally, the acetone extrography fraction shows the Soret band (~420 nm, indicated with a star), which is characteristic of porphyrins.

Table 2 shows the results for average quantification for the three developed HPTLC plates. Combined with the data shown in Figure 2, the results indicate that only \sim 9% of the Tol/THF/MeOH extrography fraction migrates up to the solvent front. For the acetone fraction, there is 53% migration. Whole asphaltenes and their Hep/Tol fraction present a similar migration of 28% and 23%, respectively. The results suggest that species in the acetone fraction are less prone to strong adsorption issues on polar surfaces such as cellulose.

Trends of Elemental Composition as a Function of Migration. HPTLC plates were analyzed by LA-ICP-MS to quantify vanadium in the eluted/non-eluted fractions. The results are presented in Figure 3 and Figure S1 and are summarized in Table 3; the results corroborate compositional similarities between whole PetroPhase 2017 asphaltenes and its Hep/Tol extrography fraction (panels A and C). For both samples, ⁵¹V is present at the deposition point (non-eluted fraction) and the solvent front (eluted species). Regarding the acetone fraction, Figure 3 B demonstrates that all vanadiumcontaining compounds migrated with the solvent front. Conversely, the Tol/THF/MeOH fraction revealed vanadium compounds that remained in the non-eluted HPTLC fraction. The quantitative results shown in Table 3 demonstrate that

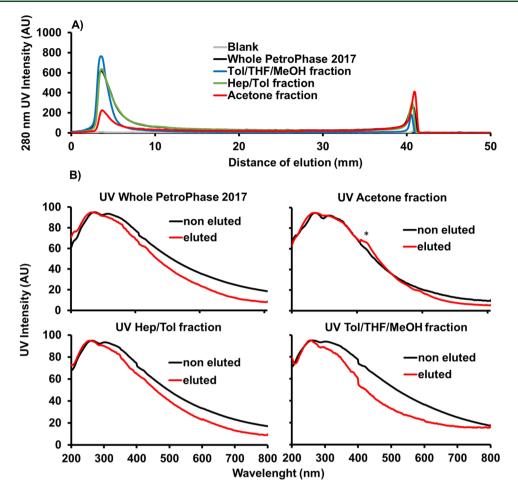


Figure 2. UV chromatograms at 280 nm (A) and UV spectra (D) and chromatograms (B) obtained by HPTLC-UV at 280 nm for all four samples and a blank.

Table 2. Results for UV Analysis at 280 nm for All Four Samples with the Average Intensity for a Total of Six Replicates and the RSD

area 280 nm	area at non- eluted point	area at eluted point	RSD at non-eluted point (%)	RSD at eluted point (%)
whole PetroPhase 2017	19000 (72%)	8000 (28%)	11.1	23.4
Tol/THF/ MeOH fraction	15000 (91.2%)	1000 (8.8%)	7.4	16.9
Hep/Tol fraction	19000 (76.9%)	6000 (23.1%)	6.6	19.8
acetone fraction	6000 (47%)	8000 (53%)	6.6	20.7

~100% of ⁵¹V-containing compounds (~20000 cps) in the acetone fraction migrated to the solvent front, whereas 100% of vanadium species in the Tol/THF/MeOH (~30000 cps) fraction remained non-eluted.

Molecular-Level Characterization by FT-ICR MS. MALDI FT-ICR MS characterization was facilitated by the use of a novel matrix designed to favor electron transfer reactions for the efficient ionization of metal complexes, such as vanadyl porphyrins.^{4,27} This MALDI matrix is a phenylenevinylene derivative that allows for greater ion abundances of porphyrin-like molecules and limits the ionization of other compounds, such as alkyl-substituted PAHs, which are also present in asphaltenes. MALDI FT-ICR spectra were collected for all samples at the deposition point (non-eluted fraction) and the solvent front (eluted fraction). Molecular formulas were assigned to the mass spectral peaks and sorted by heteroatom class (e.g., $N_4O_1V_1$ or vanadyl porphyrins), carbon number, and double bond equivalents (DBE = number of rings plus double bonds). At the sample deposition point (noneluted fraction), porphyrins were not detected in the whole asphaltene sample or in all extrography fractions. Thus, at the deposition point, vanadium-containing compounds detected by LA-ICP-MS were not ionized (as monomers) by MALDI, even when a specific matrix designed to enhance the ionization of vanadyl porphyrins was utilized. The results highlight the limitations of mass spectrometry to access the complete molecular composition of one of the most complex/ challenging samples, the asphaltenes.

Figure 4 presents isoabundance color-contoured plots of DBE versus carbon number for vanadyl porphyrins $(N_4O_1V_1 class)$ and heteroatom-substituted counterparts $(N_4O_2V_1 and N_4O_1S_1V_1 classe)$ for all eluted HPTLC fractions. In these plots, the color scale represents the relative abundance. For the $N_4O_1V_1$ class (Figure 4 A), the most abundant components (red data points) for all samples present the same carbon number (29/30) and DBE values (18). Molecular formula for DBE for all the samples are presented in Table S1. Thus, the

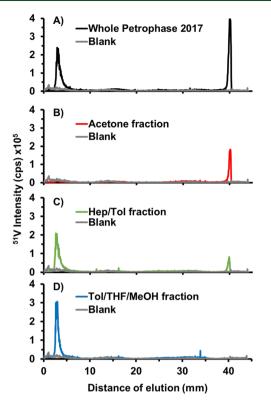


Figure 3. Chromatograms obtained by HPTLC-LA-ICP-MS analysis at *mass* 51 for all four samples and a blank.

Table 3. Results for LA-ICP-MS Analysis at *mass* 51 for All Four Samples with the Average Intensity for a Total of Six Replicates and the RSD

area ⁵¹ V	area at non-eluted point	area at eluted point	RSD at non- eluted point (%)	RSD at eluted point (%)
whole Petrophase 2017	552000 (78.9%)	148000 (21.1%)	11.4	9.8
Tol/THF/ MeOH fraction	667000 (100%)	0 (0%)	6.3	0.0
Hep/Tol fraction	500000 (95.8%)	22000 (4.2%)	2.3	31.9
acetone fraction	0 (0%)	143000 (100%)	0.0	4.1

results suggest that the PetroPhase 2017 asphaltene sample contains abundant DPEP-type porphyrins.

For the whole PetroPhase 2017 asphaltene sample, 146 molecular formulas were found; the most abundant compounds feature DBE values of 17 and 18, which indicates the dominance of Etio and DPEP porphyrins. The Hep/Tol and Tol/THF/MeOH extrography fractions only revealed 44 and 7 porphyrins species. Conversely, the acetone fraction featured the highest content of vanadyl porphyrins, as revealed by 254 molecular formulas obtained via mass spectrometry. In Figure 4 B, porphyrins with one additional oxygen atom were found only in whole asphaltenes and the acetone fraction with a 3.3-and 2.2-fold lower relative abundance than the N₄O₁V₁ class. Figure 4 C indicates that porphyrins with one sulfur atom, detected only in the acetone fraction, have abundant homologous series with DBE values from 22 to 30 and carbon

numbers between 25 and 50. The higher DBE values are consistent with S incorporated as, e.g., thiophene and benzothiophene peripheral groups.

DISCUSSION

In Figure 2 B, the spectra obtained for the non-eluted fractions for all samples are remarkably similar and are characteristic of a complex mixture of polyaromatic compounds. For the eluted HPTLC fractions, the Soret band at 420 nm was evident only in the acetone extrography fraction. As shown by Evdokimov et al.,¹⁵ porphyrins can be "invisible" in UV–vis spectroscopy due to aggregation. Thus, our results suggest that separation by HPTLC allows for the isolation of porphyrins that are not bound in asphaltene aggregates, as already proposed in our previous paper.³²

For the whole PetroPhase 2017 sample, the results obtained by UV (Figure 2) and laser ablation (Figure 3) are quite similar to our previous results.³² Indeed, nearly 80% of the vanadium remained at the deposition point (Table 3, LA-ICP-MS analysis). In that work, the Asphaltene 2017 sample was also analyzed by MALDI FT-ICR MS, but no porphyrins were detected at the deposition point. The results suggest that vanadium-containing molecules, such as porphyrins (whole asphaltenes/deposition point), are present but are not easily accessible, likely because of aggregation. These results are consistent with the UV spectroscopy behavior. Putman et al.³³ has reported limitations in MS characterization of asphaltenes due to the low ionization efficiency of specific chromatography fractions. In this work, according to Table 3 and Figure 4 A, approximately 80% of molecules containing vanadium could not be identified via ultrahigh-resolution mass spectrometry. This raises questions about what portion can be analyzed with such techniques even with previous separation steps. Furthermore, in Figure 4 B, porphyrins with one additional oxygen atom were found only in the PetroPhase 2017 sample and the acetone fraction in both cases at the eluted point with a lower relative abundance (3.34 and 2.25 times less). The UV results for the Hep/Tol fraction were quite similar (Figure 2) to those of the whole PetroPhase 2017 sample. However, according to elemental analysis, vanadium was present at a lower concentration at the eluted point (5%) for the Hep/Tol fraction than for the whole PetroPhase 2017 sample (20%). In MALDI-FT-ICR MS, the abundance of porphyrins at the eluted point was lower in the Hep/Tol fraction than in the whole sample. Collectively, the results suggest that the whole PetroPhase 2017 asphaltene sample and the Hep/Tol extrography fraction contained free and aggregate-bound porphyrins.

HPTLC-LA-ICP-MS analysis (Figure 3 D) of the Tol/ THF/MeOH extrography fraction indicates that ⁵¹V is present only at the deposition point; however, no Soret band was detected by UV-vis spectroscopy (Figure 2 B). For MALDI FT-ICR MS experiments, despite the use of the custom MALDI matrix for selective ionization of metal complexes, no porphyrins were detected at the deposition point. Thus, noneluted vanadium-containing compounds are not easily accessible and could be trapped into asphaltene aggregates that prevent their detection by UV-vis spectroscopy and as monomers in FT-ICR MS. The eluted fraction (~8.8% in UV) from Tol/THF/MeOH presents a vanadium content below the LA-ICP-MS detection limit. However, it reveals 7 N₄O₁V₁ molecular formulas; such compounds could be completely "free" or present on the aggregate surface and therefore

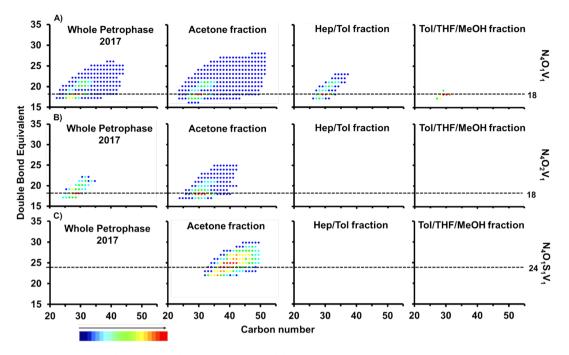


Figure 4. DBE/# maps of V1O1N4 (A), V1O2N4 (B), and V1O1N4S1 (C) for all four samples by MALDI FT-ICR MS at the eluted point.

available for MS analyses. As reported by Chacon-Patiño,²⁰ the Tol/THF/MeOH extrography fraction contains abundant multicore motifs (archipelago) and heteroatom-rich compounds. This fraction exhibits stronger aggregation trends and may comprise the most problematic asphaltene species that are likely responsible for demetalation problems in refineries.

Conversely, the acetone fraction presents the highest intensities for the three different types of porphyrins (Figure 4), and the Soret band present in the UV-vis spectrum (Figure 2) at the eluted point suggests a high amount of easily accessible porphyrins. The most abundant molecular formulas have the same carbon number and DBE values as those found for other samples (whole asphaltenes, Hep/Tol, and Tol/ THF/MeOH). These results suggest that the easily accessible porphyrins are likely the same compounds in all four samples and were concentrated in the acetone fraction, which exhibits weaker adsorption on polar surfaces (i.e., cellulose) and weaker aggregation tendencies than for the other solvents; therefore, it is likely that porphyrins were not locked into aggregates in this fraction. Furthermore, the UV results in Figure 2 A suggest that a portion of the sample (47%) remained at the deposition point, but no vanadium was found by the elemental LA-ICP-MS and molecular-level FT-ICR-MS analyses. These results suggest that Soxhlet extraction, coupled with HPTLC, allows for the separation of free porphyrins from those that are locked into asphaltene aggregates. The results indicate that the vanadium present in the acetone extrography fraction is easily accessible and constitutes free porphyrins. Chacon-Patiño et al.⁵ demonstrated that acetone fractions from geologically diverse asphaltene samples are enriched in single-core or island-type asphaltenes.²⁰ Therefore, the results indicate that vanadyl porphyrins in the acetone fraction exhibit weak intermolecular interactions with asphaltene aggregates mostly comprised of single-core motifs; HPTLC separation by cellulose plates and DCM solvent can disrupt those interactions. Conversely, vanadyl porphyrins linked to multicore-type asphaltenes in the Tol/THF/MeOH extrography fraction are strongly linked to the nanoaggregates, which precludes their elution during HPTLC separation.

CONCLUSIONS

Our previous works³² using GPC-HPTLC with LA-ICP-MS suggest that porphyrins that migrate until they reach the solvent front using DCM/MeOH were more accessible as free porphyrins than those that remained at the deposition point. Here, extrography followed by HPTLC separation was used on asphaltene samples. Then, molecular and elemental detection to analyze vanadium and vanadyl porphyrins was successfully used. The combination of molecular and elemental analyses proved that a large quantity of porphyrins remained inaccessible to molecular analyses because of matrix effects and ionization preferences and that these porphyrins are linked to archipelago asphaltenes. The MALDI FT-ICR MS analyses showed that in all the samples, porphyrins that migrate until they reach the solvent front are easier to analyze than those that remain at the deposition point. The acetone fraction showed a great quantity of easily ionizable porphyrins that migrated entirely until they reached the solvent front. These results suggest that there was a high quantity of free porphyrins or porphyrins with weak interactions with the island nanoaggregates in the acetone fraction. In contrast, the Tol/ THF/MeOH fraction, which contains a great quantity of vanadium, remains at the deposition point and gives almost no signal by MALDI-FT-ICR MS, which proves the strong interaction between porphyrins and archipelago-type nanoaggregates or an effective occlusion of the porphyrins into the nanoaggregates. These results suggest that when ultrahighresolution mass spectrometry is used on asphaltene samples, even after previously using a separation step, only a part of the molecular complex structure can be revealed. Further works should concentrate on the non-eluted part of the Hep/Tol and Tol/THF/MeOH fractions to determine the kinds of porphyrins present and their links with the asphaltene matrices.

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.energyfuels.0c02525.

(Figure S1) Six point calibration with vanadyl porphyrins after HPTLC separation and concentration obtained for the four samples by HPTLC-LA-ICP-MS; (Table S1) molecular formula, mass error, and signal-to-noise ratio for all the vanadyl porphyrins signal at DBE 18 for all four samples (PDF)

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Notes

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