

pubs.acs.org/JPCC

Dipolar Heteronuclear Correlation Solid-State NMR Experiments between Half-Integer Quadrupolar Nuclei: The Case of ¹¹B-¹⁷O

Rick W. Dorn, Alexander L. Paterson, Ivan Hung, Peter L. Gor'kov, Austin J. Thompson, Aaron D. Sadow, Zhehong Gan, and Aaron J. Rossini*

Cite This: J. Phys. Chem. C 2022, 126, 11652-11666



signal dephasing than when recoupling is applied on the ¹⁷O channel; however, short effective ¹¹B transverse relaxation time constants (T_2') hinder the acquisition of dephasing curves out to long recoupling durations. Application of REDOR recoupling to ¹⁷O spins was found to produce significant dephasing without compromising the ¹¹B T_2' . Comparison of experimental ¹¹B{¹⁷O} RESPDOR curves to those of numerical simulations enabled the ¹⁷O isotopic enrichment to be estimated. 2D ¹¹B{¹⁷O} D-HMQC spectra were recorded with either ¹¹B or ¹⁷O REDOR recoupling under a variety of radio frequency field conditions. Lastly, 2D $^{11}B{^{17}O}$ and $^{23}Na{^{17}O}$ D-HMQC spectra of an ^{17}O -enriched sodium borate glass were acquired to demonstrate the practical application of these heteronuclear correlation experiments to probe structural connectivity between two quadrupolar spins. Importantly, the high-field 2D $^{11}B^{-17}O$ D-HMQC NMR spectrum revealed two unique ^{17}O sites correlating to 4-coordinate BO₄ $({}^{[4]}B)$, which were attributed to the ${}^{[3]}B-O-{}^{[4]}B$ and ${}^{[4]}B-O-{}^{[4]}B$ bridging O atoms. The heteronuclear correlation experiments outlined here should be applicable to a variety of quadrupolar spin pairs.

INTRODUCTION

High-resolution magic-angle spinning (MAS) solid-state NMR (SSNMR) spectroscopy is a powerful technique to determine molecular structure within crystalline and amorphous solids. Dipolar-based heteronuclear correlation (HETCOR) SSNMR experiments, such as dipolar heteronuclear multiple quantum correlation (D-HMQC), transferred-echo double-resonance (TEDOR), and dipolar refocused insensitive nuclei enhanced through polarization transfer (D-RINEPT), are commonly used to probe through-space connectivity between two heteroatoms.¹⁻¹³ In MAS SSNMR spectroscopy, rotation of the sample about the magic angle (54.74°) with respect to the external magnetic field (B_0) will either partially or fully average the dipolar interaction between two dipolar-coupled spins. Therefore, heteronuclear dipolar recoupling techniques, such as rotational-echo double-resonance (REDOR),^{2,14-18} symmetry-based C-type (general formula: CN_n^v) and R-type (general formula: RN_n^v) sequences,¹⁹⁻²⁵ and so forth, are used to re-introduce the dipolar interaction during parts of the

application of dipolar recoupling on the ¹¹B channel yielded more

NMR experiment to enable through-space polarization transfer or to generate dipolar dephasing. Heteronuclear dipolar recoupling schemes are routinely applied to spin I = 1/2nuclei to recouple interactions with other spin I = 1/2 or quadrupolar nuclei (I > 1/2).^{6,7,9,11,12,26–36}

SSNMR spectroscopy of quadrupolar nuclei is a growing field; approximately 73% of all NMR-active nuclei are quadrupolar, with the majority having half-integer spin (i.e., I = 3/2, 5/2, 7/2, or 9/2). Furthermore, many materials of interest that benefit from SSNMR structural characterization, such as metal–organic frameworks,^{37–40} nanomaterials,^{41–45} oxide glasses,^{46–50} heterogenous catalysts,^{51–56} and pharma-

Received: April 20, 2022 Revised: June 20, 2022 Published: July 7, 2022





pubs.acs.org/JPCC

ceuticals,^{11,36,57-64} contain abundant quadrupolar spins. Therefore, HETCOR NMR experiments between two quadrupolar nuclei could be useful to better elucidate the chemical structure in many systems. Previously, ²⁷Al-¹⁷O 2D correlation SSNMR spectra were obtained by using HMQC experiments with coherence transfers mediated by scalar (I-) couplings.⁶⁵⁻⁶⁸ Such experiments are essentially directly analogous to the I = 1/2 case because scalar couplings are not averaged by MAS and no radio frequency (RF) pulses are required during the J-evolution periods. J-based experiments are very powerful to determine molecular structures because they directly probe through-bond structural connectivity and have high theoretical efficiencies. For optimal efficiency, Jbased experiments require one-bond J-couplings (¹J) that are approximately larger than the inverse of the refocused transverse relaxation constant $(T_2')^{.65}$ However, as shown here for ¹¹B-¹⁷O, dipolar-based experiments are much more efficient than I-based experiments because dipolar coupling constants are significantly larger than J-coupling constants (~2125 Hz compared to ~35 Hz, respectively), and ¹¹B T_2 ' are much shorter than the inverse of ${}^{1}I({}^{11}B-{}^{17}O)$.

Multiple groups have previously demonstrated that 2D dipolar HETCOR NMR spectra between two half-integer quadrupolar spins can be recorded via cross-polarization (CP) under MAS conditions.⁶⁹⁻⁷² Unfortunately, spin-locking the central-transition (CT) of a half-integer quadrupolar nucleus is often challenging and highly sample- and site-dependent.^{73,74} Studies applying heteronuclear dipolar recoupling schemes to half-integer quadrupolar spins are relatively rare and have been primarily limited to obtaining one-dimensional (1D) REDORtype spectra.^{46–48,71,75–78} To the best of our knowledge, there have only been two reports of 2D dipolar-based HETCOR NMR spectra between two half-integer quadrupolar nuclei obtained by applying heteronuclear dipolar recoupling to one of the quadrupolar spins $(^{23}Na-^{27}Al \text{ and }^{11}B-^{27}Al)$.^{79,80} In these examples, 2D ²³Na{²⁷Al}/²⁷Al{²³Na} D-HMQC and D-RINEPT spectra were acquired with either rotary resonance recoupling (R^3) , $R2_1^1$, or SR4 applied to either the ²³Na or ²⁷Al spins,⁷⁹ while 2D ${}^{11}B{}^{27}Al{}^{/27}Al{}^{11}B{}$ D-HMQC and D-RINEPT spectra were recorded with synchronous phase inversion $R^{\bar{3}}$ (SPI- R^3 ; $C2_2^1$ symmetry) applied to either the ¹¹B or ²⁷Al spins.⁸⁰ Throughout this article, the $X{Y}$ notation indicates that the X spin is directly observed and the Y spin is indirectly detected.

In this contribution, we show that REDOR recoupling can be applied to either ¹¹B (I = 3/2) or ¹⁷O (I = 5/2) to reintroduce the ${}^{11}B-{}^{17}O$ dipolar interactions and allow ${}^{11}B{}^{17}O{}$ resonance-echo saturation-pulse double-resonance (RE-SPDOR) and D-HMQC NMR experiments (Figure 1A,B, respectively). First, ¹¹B{¹⁷O} RESPDOR NMR experiments were used to confirm the successful re-introduction of the dipolar interaction and estimate the ¹⁷O-labeling percentages in ¹⁷O-enriched boric acid and benzene diboronic acid (BDBA). ¹¹B{¹⁷O} *J*-resolved experiments were also performed to determine the one-bond ¹¹B-¹⁷O J-coupling constants and estimate the ¹⁷O isotopic abundance. 2D ¹¹B{¹⁷O} D-HMQC NMR spectra were recorded by applying first-order R^3 or REDOR heteronuclear dipolar recoupling to either the ¹¹B or ¹⁷O spins. Numerical simulations of ¹¹B{¹⁷O} D-HMQC suggest that the experiments are relatively robust to both the ¹¹B and ¹⁷O quadrupolar coupling constant (C_Q) and also predict that the experiments operate with similar efficiencies at



Figure 1. (A) ${}^{11}B{}^{17}O{}$ RESPDOR and (B) ${}^{11}B{}^{17}O{}$ D-HMQC NMR pulse sequences with (C) R^3 or REDOR heteronuclear dipolar recoupling (denoted Rec.) applied to either ${}^{11}B$ or ${}^{17}O$.

a variety of MAS frequencies ($\nu_{\rm rot} = 10-50$ kHz). Lastly, 2D $^{11}B\{^{17}O\}$ and $^{23}Na\{^{17}O\}$ D-HMQC spectra of an ^{17}O -enriched sodium borate glass were acquired to demonstrate the practical utility of these heteronuclear correlation experiments to probe the structure of amorphous materials.

METHODS

Materials. BDBA (also referred to as 1,4-phenyldiboronic acid; Oakwood Chemical—ACS Reagent) was ¹⁷O-enriched via recrystallization in 39.3% $H_2^{17}O$ (Cortecnet). 45.1 mg of BDBA and 106.8 mg of 39.3% $H_2^{17}O$ were mixed in a capped vial and heated to 70 °C in an oven for *ca.* 2 days. There was *ca.* 8 mg (corresponding to *ca.* 7%) of $H_2^{17}O$ lost during the heating process. The vial was then placed in a Schlenk flask where excess water was removed at room temperature under a dynamic vacuum for *ca.* 1 day.

¹⁷O-enriched boric acid was synthesized via the hydrolysis of borane (BH₂) with 39.3% ¹⁷O-enriched H₂O (Cortecnet). An oven-dried 50 mL round bottom Schlenk flask with a microstir bar was cooled under vacuum on a Schlenk line and then filled with N2. The flask was charged with 1 M BH3•THF solution (9.7 mL, 9.7 mmol; Sigma-Aldrich) which was added using a syringe under a N2 purge. The BH3•THF solution was cooled to -78 °C in a dry-ice/acetone bath for 5 min. 39.3% $H_2^{17}O$ (0.59 mL, 31.1 mmol, 3.2 equiv) was added dropwise using a syringe over 2 min. Slight bubbling was observed. Upon full addition of H₂¹⁷O, the solution was allowed to stir for 30 min at -78 °C. The solution was then warmed to room temperature and stirred for 2 h. The volatiles were removed in vacuo overnight to yield a white powder (0.607 g), which was used without further purification. The ¹⁷O enrichment of boric acid was determined to be 30% from solution ¹⁷O NMR spectroscopy (see the Supporting Information for more details).

An ¹⁷O-enriched sodium borate glass was synthesized with a composition of $41.5Na_2O-58.5B_2O_3$ (values given in mol %). Appropriately weighed amounts of sodium carbonate (99.999% trace metals basis, Sigma-Aldrich) and 30% ¹⁷O-enriched boric acid were mixed and thoroughly ground in a

porcelain mortar and pestle. The mixed reagents were transferred to a 95/5 Pt/Au crucible which was topped with a Pt lid. The crucible and lid were placed in a 1050 °C furnace for approximately 40 min. The melt was quenched by dipping the bottom of the crucible in a room-temperature water bath. The resulting glass was clear, colorless, and free of any observable bubbles. The glass was stored in a desiccator and was ground to a powder and packed into a rotor immediately before NMR experiments.

Solid-State NMR Spectroscopy at 9.4 T. Solid-state NMR spectroscopy experiments were performed on a 9.4 T $[\nu_0(^1\text{H}) = 400 \text{ MHz}]$ Bruker wide-bore magnet equipped with a Bruker Avance III HD console and a 2.5 mm triple-resonance MAS NMR probe. ¹H chemical shifts were referenced to 1% tetramethylsilane (TMS) in CDCl₃ by using solid adamantane $[\delta(^1\text{H}) = 1.71 \text{ ppm}]$ as a secondary standard. ¹¹B, ¹⁷O, and ²³Na shifts were indirectly referenced using the previously published IUPAC-recommended relative NMR frequencies.⁸¹ The same shift referencing procedure was also used for 14.1 T NMR experiments. All NMR spectra were processed using Bruker Topspin 3.6.1 software. 2D ¹¹B multiple-quantum MAS (MQMAS) analytical simulations were performed using ssNake NMR software.⁸²

All experimental NMR parameters [magnetic field strength (B₀), MAS rate, recycle delay ($\tau_{\rm rec\ delay}$), number of scans, t_1 TD points, t_1 dwell (Δt_1), t_1 acquisition time, dipolar recoupling duration, and total experimental times] are given in Table S1. 2D ¹¹B split- t_1 MQMAS experiments were performed with previously described pulse sequences.⁸³⁻⁸ The ¹¹B 3Q excitation and 1Q reconversion pulse lengths were 3.45 and 0.98 μ s in duration, respectively, and applied at RF fields of ca. 186 kHz. 5.5° tip-angle single-pulse ¹¹B NMR experiments were performed with a 0.833 μ s pulse length, corresponding to an 8.3 kHz RF field and 16.7 kHz CT nutation frequency. ²³Na{¹⁷O} D-HMQC experiments were performed with rotor-assisted population transfer (RAPT)applied ± 250 kHz off-resonance applied to the ²³Na spins prior to the start of the experiment to enhance overall sensitivity.^{87,88} ²³Na CT-selective $\pi/2$ and π pulse lengths were 10 and 20 μ s in duration, respectively, corresponding to a 12.5 kHz RF field and 25 kHz CT nutation frequency. The ¹⁷O CTselective $\pi/2$ pulse length was 10 μ s in duration, corresponding to an 8.3 kHz RF field and 25 kHz CT nutation frequency. 100 kHz ¹H RF field SPINAL-64 heteronuclear decoupling was performed throughout all NMR experiments.⁸

Solid-State NMR Spectroscopy at 14.1 T. Solid-state NMR spectroscopy experiments were performed on a 14.1 T $[\nu_0({}^{1}\text{H}) = 600 \text{ MHz}]$ Bruker wide-bore magnet equipped with a Bruker NEO console and either a 4 mm triple-resonance MAS NMR probe (${}^{11}\text{B}{}^{17}\text{O}$) experiments) or a 2.5 mm triple-resonance MAS NMR probe configured in the double-resonance mode (${}^{17}\text{O}$ MQMAS and ${}^{11}\text{B}$ DQ-SQ experiments). The ${}^{1}\text{H}$, ${}^{11}\text{B}$, and ${}^{17}\text{O}$ channels of the 4 mm NMR probe were isolated with either bandpass (${}^{1}\text{H}$ and ${}^{11}\text{B}$) or lowpass (${}^{17}\text{O}$) filters. All NMR spectra were processed using Bruker Topspin 4.0.7 software. 2D ${}^{17}\text{O}$ MQMAS analytical simulations were performed using ssNake NMR software.⁸²

The ¹¹B CT-selective $\pi/2$ and π pulse lengths were 15 μ s and 30 μ s in duration, respectively, corresponding to an 8.3 kHz RF field and 16.7 kHz CT nutation frequency. The ¹⁷O CT-selective $\pi/2$ and π pulse lengths were 10 μ s and 20 μ s in duration, respectively, corresponding to an 8.3 kHz RF field and 25 kHz CT nutation frequency. 2D ¹¹B{¹⁷O} D-HMQC

experiments were performed with our recently described Arbitrary Indirect Dwell (AID) t_1 acquisition mode and a 25 μ s t_1 dwell.⁹⁰ RAPT was applied to the ¹¹B spins with a transmitter offset of ±550 kHz and a 10 kHz RF field prior to the start of NMR experiments.^{87,88} ¹¹B{¹⁷O} RESPDOR experiments were performed with a 1.5 rotor period saturation pulse (150 μ s for a 10 kHz MAS frequency) with a *ca.* 33 kHz ¹⁷O RF field. 2D ¹⁷O split- t_1 MQMAS experiments were performed with previously described pulse sequences.⁸³⁻⁸⁵ The ¹⁷O 3Q excitation and 1Q reconversion pulse lengths were 5.75 and 1.25 μ s in duration, respectively, and applied with RF fields of ca. 68 kHz. A 2D ¹¹B dipolar double-quantum-singlequantum (DQ-SQ) homonuclear correlation NMR spectrum of the ¹⁷O-enriched sodium borate glass was recorded with the BR2¹₂ homonuclear dipolar recoupling sequence.⁹¹ A CTselective π pulse was applied during the ¹¹B DQ t_1 evolution to ensure that only CT DQ coherences between two ¹¹B spins were observed.⁹² Either $\overline{83.3}$ kHz (¹¹B-¹⁷O experiments) or 100 kHz (¹¹B DQSQ and ¹⁷O MQMAS) ¹H RF field SPINAL-64 heteronuclear decoupling was performed throughout all NMR experiments.⁸⁹

Solid-State NMR Spectroscopy at 18.8 T. Solid-state NMR spectroscopy experiments were performed on an 18.8 T $[\nu_0(^1\text{H}) = 800 \text{ MHz}]$ mid-bore magnet equipped with a Bruker AVANCE III HD console and a 3.2 mm low-E triple-resonance MAS NMR probe. ¹⁷O chemical shifts were referenced to tap water ($\delta_{iso} = 2.83 \text{ ppm}$) with respect to D₂O ($\delta_{iso} = 0.0 \text{ ppm}$). ¹¹B shifts were indirectly referenced using the previously published IUPAC-recommended relative NMR frequency.⁸¹ All NMR spectra were processed using Bruker Topspin 3.6.1 software.

The ¹¹B CT-selective $\pi/2$ and π pulse lengths were 6 and 12 μ s in duration, respectively, corresponding to a 20.8 kHz RF field and 41.6 kHz CT nutation frequency. The ¹⁷O CT-selective $\pi/2$ and π pulse lengths were either 10 and 20 μ s or 15 and 30 μ s in duration, respectively, corresponding to either an 8.3 or 5.6 kHz RF field and a 25 or 16.7 kHz CT nutation frequency, respectively. 2D ¹¹B{¹⁷O} D-HMQC experiments were performed with our recently described AID t_1 acquisition mode and either a (boric acid) 40 μ s or (sodium borate glass) 12.5 μ s t_1 dwell.⁹⁰ RAPT was applied to the ¹¹B spins with a transmitter offset of ±500 kHz and either an 83.8 (boric acid) or 46.5 (borate glass) kHz RF field prior to the start of NMR experiments.^{87,88} 100 kHz ¹H RF field SPINAL-64 heteronuclear decoupling was performed throughout all NMR experiments on boric acid.⁸⁹

Solid-State NMR Spectroscopy at 35.2 T. 1D ¹¹B and ¹⁷O NMR spectra were recorded at 35.2 T using a 36 T seriesconnected hybrid (SCH) magnet (built and stationed at the National High Magnetic Field Laboratory in Tallahassee, FL) equipped with a Bruker AVANCE NEO console and a 2.0 mm HXY MAS NMR probe.93 11B chemical shifts were referenced to the BO4 ¹¹B NMR signal observed at 18.8 T because the quadrupolar induced shift (QIS) is negligible when $C_Q < 0.5$ MHz at high field. ¹⁷O chemical shifts were indirectly referenced through ¹¹B shifts using the previously published IUPAC-recommended relative ¹⁷O NMR frequency.⁸¹ All NMR spectra were recorded with a 20 kHz MAS frequency. Quantitative ¹¹B and ¹⁷O NMR spectra were recorded with a 15° and 30° single-pulse experiment, respectively, and RF fields of 25 kHz (50 kHz ¹¹B CT nutation frequency) and 8.3 kHz (25 kHz ¹⁷O CT nutation frequency), respectively.

¹¹B–¹⁷O RESPDOR, *J*-Resolved, and D-HMQC Error Analysis. Error bars were added to the experimental data points for all ¹¹B–¹⁷O RESPDOR and *J*-resolved curves and the ¹¹B{¹⁷O} D-HMQC efficiencies to illustrate the level of uncertainty in the measurements. For all ¹¹B{¹⁷O} RESPDOR and *J*-resolved curves, error bars are given by

error
$$(\tau) = \frac{\sqrt{S_{0,\text{norm}} + S_{\text{norm}}} \times S_{0,1}}{2 \times \text{SNR}_{0,1} \times S_{0,\text{norm}}}$$

where $S_{0,norm}$ and S_{norm} are the normalized control and dephased intensities for a given τ period (i.e., dipolar recoupling or *J*-evolution duration), respectively, $S_{0,1}$ ($S_{0,1} =$ 1) is the control point that all control and dephased points are normalized to (i.e., maximum intensity, which will be the first time point), and SNR_{0,1} is the signal-to-noise ratio of the first control point (i.e., point $S_{0,1}$). ¹¹B{¹⁷O} D-HMQC efficiency error bars were determined based on the SNR of the 1D ¹¹B{¹⁷O} D-HMQC NMR spectrum

error =
$$\frac{\sqrt{2}}{\text{SNR}}$$
 × efficiency

 $2D {}^{11}B{}^{17}O{}$ D-HMQC integral error bars were determined based on the SNR of an ${}^{17}O{}$ slice extracted at a frequency where the ${}^{11}B{}$ signal intensity was largest

error =
$$\frac{\sqrt{2}}{\text{SNR}} \times \text{integral}$$

Solid-State NMR Numerical Simulations. Numerical solid-state NMR simulations of ¹¹B{¹⁷O} RESPDOR and D-HMQC experiments were performed using SIMPSON v4.2.1.^{94–96} All simulations were performed at $B_0 = 14.1$ or 18.8 T with the rep168 crystal file, either (RESPDOR) 10 or (HMQC) 8 γ angles, and a 1 μ s maximum time duration where the Hamiltonian was considered time-independent. 2D heat plots of the ¹¹B{¹⁷O} D-HMQC efficiency as a function of ¹¹B and ¹⁷O C_Q were obtained by running 2D SIMPSON simulations in which the ${}^{11}\text{B}$ C_Q was stepped from 0.5 to 4.5 MHz in steps of 0.25 MHz for each ${}^{17}\text{O}\ \hat{C}_{Q}$ value (4–9 MHz, in steps of 0.25 MHz). The ¹¹B and ¹⁷O isotropic shifts were updated for each individual simulation such that the ¹¹B/¹⁷O center of gravity of the MAS powder pattern was at 0.0 ppm by calculating the QIS for each C_Q value. Two ${}^{11}B-{}^{17}O$ D-HMQC SIMPSON simulations were performed for each $^{11}\text{B}/^{17}\text{O}$ C_Q value, where the phase of the second ^{17}O $\pi/2$ pulse was incremented by 180°. The HMQC intensity for each ${}^{11}\text{B}/{}^{17}\text{O}$ C_Q value was obtained by subtracting the intensity of the HMQC recorded with a 180° phase inversion from that of the HMQC recorded with a 0° phase and dividing the total signal by 2 (to account for addition of "two scans"). This phase cycling procedure was performed to obtain only the HMQCfiltered signal. The ¹¹B{¹⁷O} D-HMQC efficiency was then obtained by comparing the overall D-HMQC signal intensity to the signal intensity of a ¹¹B spin echo with identical pulse parameters and spin-system parameters. The 2D heat plots were then constructed by importing the SIMPSON simulation output files into a matrix using MATLAB R2019a. 2D HMQC SIMPSON simulations used ¹¹B and ¹⁷O Euler angles of boric acid which were obtained from plane-wave density functional theory (DFT) calculations (vide infra, Figure S1). General 2D HMQC and RESPDOR SIMPSON input files, in addition to

the MATLAB codes to generate the 2D heat plots, are available in the Supporting Information.

Periodic Plane-Wave DFT Calculations. The gaugeincluding projector-augmented wave (GIPAW)⁹⁷ approach as implemented in CASTEP version 2017 T2⁹⁸ was used for all periodic plane-wave DFT calculations. Only H atoms, if applicable, were geometry-optimized, while all other atomic coordinates were fixed to the positions determined from X-ray diffraction. Geometry optimizations of the H atom positions, when applicable, were converged to an energy threshold of 5.0 \times 10⁻⁶ eV/atom, a maximum force threshold of 0.01 eV/Å, and a maximum displacement threshold of 5.0×10^{-4} Å using the Broyden-Fletcher-Goldfarb-Shanno geometry optimization method.⁹⁹ All calculations utilized the generalized gradient approximation with the Perdew-Burke-Ernzerhof exchangeapproximation with the Perdew-Durke-Entreent exchange-correlation functional,¹⁰⁰ Tkatchenko–Scheffler (TS) disper-sion corrections,¹⁰¹ on-the-fly ultrasoft pseudopotentials (default settings in CASTEP),^{102,103} zeroth-order regular approximation relativistic treatments,¹⁰⁴ a 0.07 Å⁻¹ k-point spacing, and a 630 eV kinetic energy cutoff. Select DFT calculations were also performed with a 0.03 $Å^{-1}$ k-point spacing to confirm that the ¹⁷O isotropic shieldings (σ_{iso}) converged with a 0.07 $Å^{-1}$ k-point spacing (the difference in shielding was between 0.00 and 0.04 ppm). DFT-calculated ¹⁷O $\sigma_{\rm iso}$ were converted to ¹⁷O isotropic chemical shifts ($\delta_{\rm iso}$) via a previously published calibration plot utilizing identical DFT calculation parameters.³⁴ All crystal files (.cif) and NMR calculation output files (.magres) are available in the Supporting Information.

RESULTS AND DISCUSSION

¹¹B{¹⁷O} RESPDOR. ¹¹B-¹⁷O dipolar HETCOR SSNMR experiments were performed on BDBA and boric acid ¹⁷Olabeled to ca. 10% (determined via ¹¹B{¹⁷O} J-resolved) and 30% (determined via solution ¹⁷O NMR), respectively, via either recrystallization in water (BDBA) or hydrolysis of BH₃ (boric acid). The ¹⁷O isotopic enrichment of the water used in synthesis was 39.3% in both cases. Molecular structures of boric acid and BDBA are shown in Figure S1. ${}^{11}B{}^{17}O{}$ RESPDOR NMR experiments were initially performed to confirm the successful re-introduction of the ${}^{11}B-{}^{17}O$ dipolar interaction under MAS (Figure 1A,C). As mentioned above, REDOR-type experiments have been performed for several quadrupolar nucleus combinations $({}^{11}B-{}^{17}O, {}^{11}B-{}^{23}Na, {}^{11}B-{}^{27}Al$, and ${}^{23}Na-{}^{51}V$). ${}^{46-48,71,75-78}$ We note that a RESPDOR experiment performed with REDOR dipolar recoupling is essentially the same as the rotational-echo adiabatic-passage double-resonance (REAPDOR) experiment introduced by Gullion and co-workers.^{26-28,30}

In a ¹¹B{¹⁷O} RESPDOR experiment (Figure 1A), increasing the recoupling duration converts a larger fraction of in-phase ¹¹B magnetization (e.g., I_x) to non-detectable, antiphase ¹¹B-¹⁷O magnetization [e.g., $2I_yS_z \sin(\pi\kappa\omega_D\tau)$]. The anti-phase magnetization is not refocused when a saturation pulse is applied to the ¹⁷O spins, causing a loss of the signal (dephasing). The rate at which the ¹¹B NMR signal dephases is dependent on the strength of the dipolar coupling frequency which is proportional to the gyromagnetic ratio (γ) of both nuclei and the inverse cube of their inter-nuclear distance $\left(\omega_D \propto \frac{\gamma\gamma_i}{r_s^2}\right)$. Note that ω_D depends upon the orientation of the

dipolar vector with respect to the external magnetic field. The

requirement of generating anti-phase magnetization in a RESPDOR experiment serves as an ideal test if the dipolar interaction between two quadrupolar spins can be efficiently recoupled.

¹¹B $\{^{17}O\}$ RESPDOR curves of boric acid and BDBA were recorded with $B_0 = 18.8$ T and 14.1 T, respectively, and a 10 kHz MAS frequency (Figure 2A,B). REDOR heteronuclear



Figure 2. (A,B) ¹¹B{¹⁷O} RESPDOR curves of (A) boric acid and (B) BDBA recorded with either (green) 8.3 kHz RF field REDOR recoupling applied to the ¹⁷O spins or (orange) 6.25 kHz RF field REDOR recoupling applied to the ¹¹B spins. (C) ¹¹B T_2' measurement of boric acid under (orange) ¹¹B or (green) ¹⁷O dipolar recoupling. (D) ¹¹B{¹⁷O} *J*-resolved curve of boric acid recorded with a CT-selective ¹⁷O inversion (π) pulse. Experimental data points are shown as circles, and (A,B) numerical simulations or (C,D) analytical fits are shown as solid lines. Numerical simulations used a 2125 Hz ¹¹B-¹⁷O dipolar coupling constant. NMR experiments were performed at either (A,C,D) 18.8 T or (B) 14.1 T with a 10 kHz MAS frequency.

dipolar recoupling was applied either to the ¹⁷O spins with an 8.3 kHz RF field (CT nutation frequency = 25 kHz) or to the ¹¹B spins with a 6.25 kHz RF field (CT nutation frequency = 12.5 kHz). The ¹¹B{¹⁷O} RESPDOR curves recorded with REDOR applied to the ¹¹B spins yielded more signal dephasing and a faster build-up to maximum dephasing than when REDOR was applied to the ¹⁷O spins. The faster build-up to maximum dephasing than when REDOR was applied to the ¹¹B CT spins to ¹⁷O spins residing in both CT spin states ($m_1 = \pm 1/2$) and satellite transition (ST) spin states ($m_1 = \pm 3/2, \pm 5/2$). In comparison, CT-selective REDOR recoupling π pulses applied to ¹⁷O likely

only recouple the ¹⁷O spins residing in CT states effectively, leading to a slower build-up and lower extent of dephasing. Although applying recoupling to the observed ¹¹B spins yields faster build-up and a higher extent of dephasing, the intensity of the control signal (S_0) more quickly reaches zero due to a decrease in the ¹¹B T_2' (Figure 2C). Hence, the acquisition of curves out to longer durations of heteronuclear dipolar recoupling is challenging with ¹¹B recoupling. In addition, when ¹¹B REDOR recoupling is applied to boric acid, the dephasing curve $(1 - S/S_0)$ tends toward zero at longer durations of recoupling (Figure 2A, orange). The decrease in the dephasing curve observed at longer durations of recoupling likely arises because fluctuations in the experimental MAS frequency will result in improper refocusing of the ¹¹B quadrupolar interaction and chemical shift anisotropy. 11B recoupling may also partially scramble the spins between all states/coherences and will also recouple ¹¹B homonuclear dipolar couplings between CT and ST ¹¹B spins. Consistent with our results, Zheng and co-workers recently showed that the T_2' is significantly reduced when dipolar recoupling is applied to the directly detected channel in ¹¹B-²⁷Al D-HMQC experiments.⁸⁰ A similar trend in ¹¹B T_2' was observed for \dot{BDBA} (~5.5 ms under ^{17}O recoupling and ~1.5 ms under ^{11}B recoupling, Figure S2A).

Applying REDOR recoupling to the ¹⁷O spins enabled application of longer durations of heteronuclear dipolar recoupling without any clear decrease in the extent of dephasing at longer recoupling durations (Figure 2A, green). The observation that recoupling on the indirectly detected channel is more robust to MAS fluctuations is consistent with prior work from Garbow and Gullion, where they showed that π pulse mistiming of a few microseconds leads to significant intensity losses when applying REDOR recoupling to the observed spins.¹⁰⁵ In summary, based on the longer ¹¹B T_2' and robustness to MAS fluctuations, application of REDOR recoupling to the indirectly detected spins (i.e., ¹⁷O) is generally preferred for RESPDOR-type experiments.

Measuring ¹⁷O Isotopic Enrichment. From the X-ray crystal structures of BDBA and boric acid, it is known that the B–O bond length is *ca.* 1.35 Å.^{106,107} Therefore, by scaling the intensity of two-spin (i.e., ¹¹B-¹⁷O) SIMPSON numerically simulated ¹¹B{¹⁷O} RESPDOR curves to match a B-O internuclear distance of 1.35 Å (dipolar coupling constant = 2125 Hz), the ¹⁷O-labeling percentage can be estimated. This is because the maximum dephasing of the ¹¹B NMR signal is dependent on the ¹⁷O isotopic abundance (see the Supporting Information for more discussion). Following this procedure, 17 O-labeling percentages were determined with $^{11}B{^{17}O}$ RESPDOR to be ca. 30-33 and 8-13% for boric acid and BDBA, respectively (Figure 2A,B). The estimated ¹⁷O isotopic enrichment of 30-33% determined for boric acid matches well with that determined from solution ¹⁷O NMR spectroscopy (~30%, Figure S3). We note that the experimental ${}^{11}B{}^{17}O{}$ RESPDOR curves recorded with REDOR recoupling applied to the ¹¹B spins show higher than expected dephasing (before decreasing due to MAS fluctuations), which is likely due to recoupling of intermolecular ¹¹B-¹⁷O spin pairs from nearby molecules within the crystal lattice.

The ¹⁷O isotopic abundance was also estimated via ¹¹B{¹⁷O} *J*-resolved NMR experiments recorded with a CT-selective ¹⁷O inversion (π) pulse to selectively invert only the ¹⁷O CT (Figures 2D and S2B). Analytical fits that take into account the statistical probability of having an attached ¹⁷O atom residing



Figure 3. 2D ¹¹B{¹⁷O} D-HMQC spectra of (A–D) boric acid and (E,F) BDBA recorded at either (A,B) $B_0 = 18.8$ or (C–F) 14.1 T with a 10 kHz MAS frequency and 0.8 ms of total heteronuclear dipolar recoupling. The heteronuclear dipolar recoupling schemes, experimental times, and recycle delays (relative to T_1) are listed within the NMR spectra. The 2D contour levels and spectral processing parameters within a set (i.e., A,B, C,D, or D–F) are identical. ¹¹B and ¹⁷O spin echo NMR spectra are displayed above the 2D projections. Molecular structures of boric acid and BDBA are shown within the NMR spectra.

in the CT spin state suggest that the ¹⁷O isotopic abundance of boric acid and BDBA is 30 and 10%, respectively, in excellent agreement with solution NMR and the ¹¹B{¹⁷O} RESPDOR results (see the Supporting Information for the J-resolved fitting procedure). The one-bond ¹¹B-¹⁷O J-coupling constants $({}^{1}I_{BO})$ were determined to be 35 Hz, similar to those of B-O containing small molecules previously observed via solution NMR spectroscopy.¹⁰⁸ Euler angles used in the numerically simulated RESPDOR curves were obtained from periodic plane-wave DFT calculations of BDBA or boric acid (Figure S1); however, the Euler angles had a minimal effect on the simulated RESPDOR curves (Figure S4A). Numerical simulations also suggest that the rate and extent of signal dephasing in a ¹¹B{¹⁷O} RESPDOR experiment performed with ¹¹B recoupling are nearly identical to those of a ${}^{13}C{}^{17}O{}$ RESPDOR experiment with REDOR recoupling applied to the ¹³C spins, suggesting that both experiments exhibit similar mechanisms for dipolar recoupling, that is, recoupling is driven by rotor-synchronized inversion or refocusing of the CT of the coupled or observed spin, respectively (Figure S4B).

2D ¹¹B{¹⁷O} **D-HMQC Experiments.** 2D ¹¹B{¹⁷O} D-HMQC NMR spectra of boric acid and BDBA were recorded at either $B_0 = 14.1$ or 18.8 T, respectively, with a 10 kHz MAS frequency and first-order R^3 (¹¹B only) or REDOR heteronuclear dipolar recoupling applied to the ¹¹B or ¹⁷O spins (Figures 3 and S5–S10). RAPT was applied to the ¹¹B spins at the beginning of all HMQC experiments (Figure 1B).^{87,88} We note that application of RAPT to the ¹⁷O spins concurrent with ¹¹B recoupling could theoretically improve sensitivity and the rate of coherence build-up;^{109,110} however, this strategy was not implemented here as the ¹⁷O channel of the NMR probe at 14.1 T exhibited a high *Q*, preventing efficient off-resonance ¹⁷O saturation.

2D ¹¹B{¹⁷O} D-HMQC spectra of boric acid (30% ¹⁷O enrichment, ¹¹B $T_1 \sim 7-10$ s) were acquired in *ca*. 0.7 or 1.4 h at $B_0 = 18.8$ T or 14.1 T, respectively, while 2D ¹¹B{¹⁷O} D-HMQC spectra of BDBA (10% ¹⁷O enrichment, ¹¹B $T_1 \sim 10$ s) were acquired in *ca*. 4 h at $B_0 = 14.1$ T. All 2D ¹¹B{¹⁷O} D-HMQC spectra showed the expected correlations between ¹¹B and ¹⁷O (Figure 3). The ¹¹B projections extracted from the 2D



Figure 4. (A–C) Experimental ¹¹B{¹⁷O} D-HMQC efficiencies for (A,B) boric acid and (C) BDBA. Efficiencies were determined by comparing 1D ¹¹B{¹⁷O} D-HMQC NMR signal integrals to those of a ¹¹B spin echo spectrum. (D–F) Relative 2D ¹¹B{¹⁷O} D-HMQC contour integrations for (D,E) boric acid and (F) BDBA. First-order R^3 or REDOR heteronuclear dipolar recoupling schemes were applied to either the ¹¹B or ¹⁷O spins (¹¹B R^3 = blue, ¹¹B REDOR = orange, and ¹⁷O REDOR = green). NMR spectra were recorded at either (A,D) 18.8 T or (B,C,E,F) 14.1 T with a 10 kHz MAS frequency and 0.8 ms of total heteronuclear dipolar recoupling.

HETCOR NMR spectra match relatively well with their ideal quadrupolar powder patterns, illustrating the ability to extract accurate ¹¹B C_Q and η values (Figures S11–S13). The breadth of the extracted ¹⁷O projections matches well with the breadth of their ideal quadrupolar powder patterns, suggesting that the magnitude of ¹⁷O C_0 can be accurately determined. However, the shape of the extracted ¹⁷O projections resembles that of η = 0, limiting the ability to accurately determine the ¹⁷O η from the 2D NMR spectra (Figures S11-S13). We also recorded a 2D ¹¹B{¹⁷O} J-HMQC NMR spectrum of boric acid at $B_0 =$ 18.8 T which reveals similarly distorted ¹¹B and ¹⁷O projections, suggesting that the dipolar recoupling sequences are not necessarily the origin of the lineshape distortions (Figures S6D and S11). The 2D ¹¹B{¹⁷O} J-HMQC spectrum required a significantly longer acquisition time to acquire a spectrum with a similar SNR as the 2D D-HMQC NMR spectra. All 2D ¹¹B{¹⁷O} D-HMQC spectra display a diagonallike contour pattern: ¹¹B low-frequency horn—¹⁷O lowfrequency horn and ¹¹B high-frequency horn—¹⁷O highfrequency horn. In both boric acid and BDBA, plane-wave DFT calculations predict that the V_{33} component of the ¹¹B electric field gradient (EFG) tensor is perpendicular to both the ${}^{11}B-{}^{17}O$ dipole vector and V_{33} component of the ${}^{17}O$ EFG tensor (β Euler angle with respect to ¹¹B EFG ~ 90°).

The overall ¹¹B $\{^{17}O\}$ D-HMQC NMR efficiencies were determined by comparing 1D HMQC spectra to a ¹¹B spin echo NMR spectrum where the total ¹¹B echo duration in both experiments was the same to account for ¹¹B T_2' relaxation (0.8 ms, Figures 4A–C and S14). Note that the ¹¹B T_2' is significantly shortened when REDOR recoupling is applied to the ¹¹B spins, further reducing the observed HMQC efficiencies (Figures 2C and S2A). For boric acid, ¹¹B $\{^{17}O\}$

D-HMQC efficiencies were *ca.* 1.7–6.3% at $B_0 = 14.1$ or 18.8 T (Figure 4A,B). At 18.8 T, application of REDOR recoupling to the 17 O spins yielded the greatest HMQC efficiency (6.3%), while at 14.1 T, application of REDOR recoupling to the ¹¹B spins was most efficient (6.3%). However, at both magnetic field strengths, the difference in HMQC efficiencies only varied by ca. 1% between recoupling on the ¹¹B or ¹⁷O channels. Numerical simulations (two-spin system) of ¹¹B{¹⁷O} D-HMQC with ¹¹B REDOR recoupling suggest an efficiency of ca. 10% as compared to a ¹¹B spin echo, which is only slightly lower than that for ¹³C{¹⁷O} D-HMQC with ¹³C REDOR recoupling (ca. 14%, Figure S4C). Considering that the overall probability of having an attached ¹⁷O to ¹¹B in 30% ¹⁷Oenriched boric acid is ~66%, the experimental ${}^{11}B{}^{17}O{}$ D-HMQC experiments operate with efficiencies similar to those predicted by numerical SIMPSON simulations. Similar trends were observed for ¹¹B{¹⁷O} D-HMQC experiments on BDBA (10% ¹⁷O enrichment) at $B_0 = 14.1$ T, where efficiencies were ca. 1.1-2.5% (Figure 4C). The decrease in HMQC efficiency for BDBA as compared to boric acid is due to lower ¹⁷Oenrichment levels.

Signal integrations over the 2D contour plots were performed to better assess the relative D-HMQC efficiencies (Figure 4D–F). The trends in the 2D signal integrations with respect to the different recoupling sequences/conditions match nearly identically to the trends observed in the 1D ¹¹B{¹⁷O} D-HMQC efficiencies. Importantly, the relative 2D signal integrations are similar regardless of which spin REDOR recoupling was applied to. This observation is important because it implies that 2D D-HMQC experiments should be feasible for many different types of quadrupolar spin pairs.



Figure 5. SIMPSON numerically simulated heat plots comparing the effects of ¹¹B and ¹⁷O C_Q on the overall ¹¹B $\{^{17}O\}$ D-HMQC efficiency under REDOR heteronuclear dipolar recoupling applied to either the (upper) ¹¹B or (lower) ¹⁷O spins with the indicated RF fields. All simulations were performed with $B_0 = 14.1$ T, a 10 kHz MAS frequency, a 2125 Hz ¹¹B-¹⁷O dipolar coupling constant, and 0.8 ms of total heteronuclear dipolar recoupling. The white π symbols indicate the ¹¹B $\{^{17}O\}$ D-HMQC efficiency for a given set of ¹¹B and ¹⁷O C_Q values.

The effect that the $^{11}\mathrm{B}$ and $^{17}\mathrm{O}$ C_{Q} have on the overall ¹¹B{¹⁷O} D-HMQC efficiencies was numerically investigated using the SIMPSON program.⁹⁴ Simulations of ¹¹B{¹⁷O} D-HMQC NMR experiments were performed with either 6.25 kHz, 12.5 kHz, or 25 kHz RF field REDOR recoupling applied to the ¹¹B spins or 8.3 kHz, 16.7 kHz, or 33.3 kHz RF field REDOR recoupling applied to the ¹⁷O spins (Figure 5). The 6.25 kHz RF field for ¹¹B REDOR recoupling is essentially the lowest achievable RF field before the recoupling becomes R^3 (5) kHz RF field). Numerical simulations were performed with B_0 = 14.1 T, a 10 kHz MAS frequency and a two-spin ${}^{11}B-{}^{17}O$ spin system with a 2125 Hz dipolar coupling constant (corresponding to a 1.35 Å internuclear distance). ${}^{11}B{}^{17}O{}$ D-HMQC efficiencies were determined by comparing numerically simulated ${}^{11}B{}^{17}O{}$ D-HMQC signal intensities to that of a ¹¹B spin echo with the same spin system (Figure S15). The ¹¹B and ¹⁷O transmitters were placed on resonance with the center of gravity of each MAS powder pattern in all simulations (see Methods).

Comparison of the numerical simulations suggests that application of REDOR dipolar recoupling to the ¹¹B spins is the most efficient recoupling scheme, in agreement with the experimental data discussed above (Figure 5). However, the numerical simulations do not account for T_2' relaxation, which explains why ¹⁷O recoupling offered similar efficiencies in the experiments. In general, the ¹¹B{¹⁷O} D-HMQC experiments are predicted to be relatively robust to both the ¹¹B C_Q (ranging from 0.5 to 4.5 MHz) and ¹⁷O C_Q (ranging from 4.0 to 0.9 MHz). However, the overall HMQC signal intensity does decrease with increasing ¹⁷O C_Q . Some of the smaller ¹¹B C_Q values also show lower D-HMQC efficiency, likely because the ¹¹B recoupling and spin echo π -pulses are no longer operating in the CT-selective regime. Additional ¹¹B{¹⁷O} D-HMQC numerical simulations were also performed with a 25

kHz or 50 kHz MAS frequency (Figure S16). Interestingly, the overall HMQC efficiency is predicted to increase at faster MAS frequencies when REDOR recoupling is applied to the ¹⁷O spins but is relatively constant at all MAS frequencies under ¹¹B recoupling. The prediction that ¹¹B{¹⁷O} D-HMQC experiments are feasible at a range of MAS frequencies further suggests the broad applicability of these experiments. Lastly, we numerically investigated the effect that the offset of the REDOR recoupling pulses has on the overall ¹¹B{¹⁷O} D-HMQC efficiency for a variety of RF fields (Figure S17). As expected, the tolerance to offset increases with the increasing RF field.

¹⁷O-Labeled Sodium Borate Glass. The practical utility of the dipolar heteronuclear correlation experiments between two quadrupolar spins was demonstrated by recording 2D ¹¹B{¹⁷O} and ²³Na{¹⁷O} D-HMQC spectra of a ¹⁷O-labeled sodium borate glass with a composition of 41.5Na2O-58.5B₂O₃ (amounts given in mol %). The ¹⁷O enrichment of the glass is estimated to be 15-20% because 30% ¹⁷O-enriched boric acid was used in the synthesis. A ¹¹B MQMAS spectrum of the sodium borate glass shows two BO3 ([3]B) ¹¹B NMR signals that were previously assigned to $B(OB)_3$ ($\delta_{iso} = 19.6$ ppm, $C_Q = 2.6 \text{ MHz}$, $\eta = 0.3$) and $B(OB)_x(O^-)_{2-x}$ ($\delta_{iso} = 18.5 \text{ ppm}$, $C_Q = 2.5 \text{ MHz}$, $\eta = 0.6$) containing one or two non-bridging O atoms (Figure S18).^{111–115} The observation that the site with a higher $\delta_{iso}(^{11}B)$ exhibits a lower η is consistent with results previously reported by Stebbins and co-workers for a glass of similar composition.¹¹¹ There is also an intense, sharp ¹¹B NMR signal near 0 ppm that can be assigned to BO₄ $(^{[4]}B)$ species. 1D¹¹B NMR spectra recorded at $B_0 = 9.4$, 18.8, or 35.2 T further confirmed the identification of all ¹¹B sites (Figure S17D).

Feller and co-workers previously described expressions for predicting the population of allowed borate units for different Table 1. Predicted Populations of Borate Units and Experimentally Determined Populations of Boron Species in the Sodium Borate Glass

Borate Units	Labels	Predicted Populations (%) ^a	Boron Species	Experimentally Determined Populations (%) ^b
Triborate	$BO_3 = T^3$	15		
B-0,0- 0, B-0 0, 0-	$BO_4 = T^4$	5	O	20
Metaborate 0^{\ominus} 0^{\ominus} 0^{-B} 0^{-B} 0^{-C}	$BO_3 = M^3$	25	O ^O	41
Diborate	$BO_3 = D^3$	15		
	$BO_4 = D^4$	15	<u>`o</u>	
	$BO_4 = L^4$	25	0, ^B , '0'	39

^{*a*}Populations were predicted based on the work of Feller et al. for $0.7 \le R < 1.0$, where $R = \text{mol } \% \text{ Na}_2\text{O/mol } \% \text{ B}_2\text{O}_3$.¹¹⁶ ^{*b*}Determined from the quantitative ¹¹B NMR spectra.

glass compositions; the glass composition is given by the parameter $R = \text{mol }\% \text{ Na}_2 \text{O/mol }\% \text{ B}_2 \text{O}_3^{.116} R = 0.71$ for the 41.5Na}_2O-58.5B}_2O_3 sodium borate glass used in this study. Following their expression for the region of $0.7 \leq R < 1.0$, the sodium borate glass is predicted to contain 20% triborate, 25% metaborate, 30% diborate, and 25% loose BO₄ units (Table 1, see the Supporting Information for more details). The predicted population of ^[4]B (45%) is slightly higher than that determined experimentally (39%). Consequently, the predicted population of metaborate units (25%) is lower than that determined experimentally (40%). Nevertheless, the experimentally determined populations of ^[3]B and ^[4]B (61 and 39%, respectively) are in reasonable agreement with theoretical predictions (55 and 45%, respectively).

Comparison of 1D ¹⁷O direct excitation NMR spectra recorded at either $B_0 = 14.1$, 18.8, or 35.2 T with a 2D ¹⁷O MQMAS spectrum recorded at $B_0 = 14.1$ T reveals two main groups of signals centered at *ca*. 57 ppm ($C_Q = 4.8$ or 5.8 MHz) and 82 ppm in the 35.2 T spectrum ($C_Q = 4.8$ MHz; Figures 6B and S19). Due to the significant CT quadrupolar line narrowing provided at 35.2 T, each group of ¹⁷O NMR signals required two sites to analytically simulate the experimental 35.2 T ¹⁷O NMR spectrum (Figures 6B and S19C and Table S2, see the Supporting Information for more discussion). Previous 2D ¹⁷O MQMAS NMR spectra of a similar sodium borate glass identified two ¹⁷O NMR signals with is otropic shifts of *ca*. 56 ppm ($P_Q = C_Q\sqrt{1 + \frac{\eta^2}{3}} = 5.6$ MHz) and 82 ppm ($P_Q = 5.0$ MHz) which were assigned to non-bridging B–O⁻ and bridging B–O–B species, respectively.¹¹¹ We note that ¹⁷O

SSNMR has been used to investigate other borate systems (sodium borate with more sodium, barium borate, and crystalline NaBO₂), and in all cases, the bridging O atoms exhibit ¹⁷O isotropic chemical shifts greater than 80 ppm.¹¹¹ Specifically, for crystalline NaBO₂, bridging and non-bridging O atoms were determined to have isotropic chemical shifts of *ca.* 92 and 80 ppm, respectively.¹¹¹

A 2D ¹¹B{¹⁷O} D-HMQC NMR spectrum of the sodium borate glass was recorded at $B_0 = 18.8$ T with a 10 kHz MAS frequency and 12.5 kHz RF field REDOR recoupling applied to the ¹¹B spins (Figure 6A). This 2D NMR spectrum was acquired in just under 1 day, illustrating that HETCOR experiments between two quadrupolar spins should be feasible for a variety of materials. The 2D¹¹B{¹⁷O} D-HMQC spectrum displays multiple correlations between ¹⁷O and ¹¹B. The ^[3]B ¹¹B NMR signals centered at *ca.* 17 ppm mainly show correlations to ¹⁷O NMR signals centered at *ca.* 75 ppm which were previously assigned to bridging O atoms ($^{[3]}B-$ O $-^{[3]/[4]}B$; Figure 6B, projection over $^{[3]}B$). There are additional, weak correlations to the ¹⁷O NMR signals centered at ca. 45 ppm which were previously assigned to non-bridging O atoms $({}^{[3]}B-O^{-};$ the center of the signal is much less than the isotropic chemical shift of 65 ppm due to quadrupolar signal broadening and the QIS). Interestingly, the ^[4]B ¹¹B NMR signal shows strong correlations to both ¹⁷O NMR signals centered around 45 and 75 ppm (Figure 6B, projection over ^[4]B). The correlation between ^[3]B and the ¹⁷O NMR signal centered at 75 ppm arises from ^[3]B $-O^{-[4]}B$ units. The correlation of ^[4]B to the ¹⁷O NMR signal centered at 45 ppm is intriguing. It is perhaps unexpected that ^[4]B contains a nonbridging O atom because the ^[4]B unit would then formally possess a net charge of -2. However, as discussed above, Feller



Figure 6. (A) 2D ¹¹B{¹⁷O} D-HMQC spectrum of an ¹⁷O-enriched sodium borate glass recorded at $B_0 = 18.8$ T with a 10 kHz MAS frequency and 12.5 kHz RF field REDOR heteronuclear dipolar recoupling applied to the ¹¹B spins. (Purple) Direct excitation ¹¹B and ¹⁷O NMR spectra are overlaid above the 2D projections. (B) Comparison of (upper) direct excitation ¹⁷O NMR spectra ($B_0 = 35.T$ or 18.8 T) to the ¹⁷O NMR spectra obtained from projections from the 2D D-HMQC spectrum: (middle) projection over ^[3]B ¹¹B NMR signals (~22–12 ppm) and (lower) projection over ^[4]B ¹¹B NMR signals (~2–0 ppm). Experimental and analytically simulated spectra are shown as (solid) black and (dashed) red lines, respectively. (C) 2D ¹¹B dipolar DQ-SQ homonuclear correlation NMR spectrum recorded at $B_0 = 14.1$ T with a 25 kHz MAS frequency and 320 μ s of total BR2¹ homonuclear dipolar recoupling. (D) Comparison of (lower) periodic plane-wave DFT-calculated ¹⁷O isotropic chemical shifts of the sodium borate glass determined experimentally.

and co-workers predicted abundant diborate species for this glass composition, and the diborates contain a $^{[4]}B-\mathbf{O}-^{[4]}B$ unit.

To further investigate the correlation between ^[4]B and the ¹⁷O NMR signal centered at 45 ppm, we recorded a 2D ¹¹B dipolar DQ-SQ homonuclear correlation NMR spectrum at B_0 = 14.1 T (Figure 6C). Intense [4]B-[4]B homonuclear correlations were observed, implying that ^[4]B units linked via bridging O atoms must be common. Abundant ^[4]B-O-^[4]B motifs were recently identified in borosilicate glasses via ¹¹B DQ-SQ homonuclear correlation NMR experiments.¹¹⁷ Periodic plane-wave DFT NMR calculations of crystalline boron oxide-based systems containing either ${}^{[3]}B-O-{}^{[3]}B$, $[^{3}]B-O-[^{4}]B$, or $[^{4}]B-O-[^{4}]B$ suggest that $[^{4}]B-O-[^{4}]B$ bridging O atoms generally exhibit a lower ¹⁷O isotropic chemical shift as compared to ${}^{[3]}B-O-{}^{[3]}B$ or ${}^{[3]}B-O-{}^{[4]}B$ bridging O atoms (Figure 6D and Table S3). Therefore, we attribute the observed ¹¹B-¹⁷O heteronuclear correlations between ^[4]B and the ¹⁷O NMR signals centered at 45 ppm to [4]B-O-[4]B units. As mentioned above, two O sites were required to analytically simulate the high-resolution 35.2 T ¹⁷O NMR spectrum for the group of signals previously assigned to $^{[3]}B-O^-$ (centered at *ca.* 52 ppm at 35.2 T). Assuming that ^[4]B-O-^[4]B exhibits a similar ¹⁷O C_Q as the other bridging O atoms (${}^{[3]}B-O-{}^{[3]/[4]}B$, 4.8 MHz), the two sites likely

correspond to ^[3]B-O⁻ (blue fit; $\delta_{iso} = 65$ ppm and $C_Q = 5.8$ MHz) and ^[4]B-O-^[4]B (brown fit; $\delta_{iso} = 47$ ppm and $C_Q = 4.8$ MHz, Table S2). The ¹⁷O NMR signals of the ^[3]B-O⁻ and ^[4]B-O-^[4]B sites are only partially resolved at 35.2 T and overlap at lower magnetic fields (i.e., $B_0 < 18.8$ T). The population of each type of O atom was predicted by scaling the calculated population of the allowed borate units discussed above by the number of O atoms for each group (see the Supporting Information for more discussion). This procedure predicts a glass containing 87% ^[3]B-O-^{[3]/[4]}B, 5% ^[4]B-O-^[4]B, and 8% ^[3]B-O⁻, in excellent agreement with populations determined from the fits of the 35.2 T ¹⁷O NMR spectrum (Table 2).

Lastly, a 2D ²³Na{¹⁷O} D-HMQC NMR spectrum was recorded at $B_0 = 9.4$ T with a 25 kHz MAS frequency and 16.7 kHz RF field REDOR applied to the ¹⁷O spins to illustrate the general applicability of these D-HMQC experiments to probe structural connectivity between two quadrupolar spins (Figure S20). All ¹⁷O NMR signals show correlations to ²³Na. However, limited structural information can be obtained from the ²³Na{¹⁷O} D-HMQC spectrum due to the disordered nature of the glass and the low spectral resolution obtained at $B_0 = 9.4$ T.

Table 2. Predicted and Experimentally Determined Population of Oxygen Species in the Sodium Borate Glass

oxygen species	predicted populations (%) ^a	experimentally determined populations (%) ^b
^[3] B-O- ^{[3]/[4]} B	87	87
$^{[4]}B-O-^{[4]}B$	5	6
^[3] B-O ⁻	8	7

^{*a*}Populations were predicted by scaling the predicted boron unit populations in Table 1 by the number of oxygen atoms for a given group. ¹¹⁶ ^{*b*}Determined from fitting of the quantitative 35.2 T ¹⁷O NMR spectrum.

CONCLUSIONS

Here, we showed that application of REDOR heteronuclear dipolar recoupling to either the ¹¹B or ¹⁷O spins can successfully re-introduce the ¹¹B-¹⁷O dipolar interaction in ¹¹B{¹⁷O} RESPDOR and D-HMQC solid-state NMR experiments. ¹¹B{¹⁷O} RESPDOR experiments were initially performed on boric acid and BDBA ¹⁷O-labeled to ca. 30% and 10%, respectively. Application of REDOR recoupling to the ¹¹B spins yielded more ¹¹B signal dephasing than ¹⁷O recoupling; however, only a few points in the RESPDOR curve could be obtained due to short ¹¹B T_2' under ¹¹B recoupling. Therefore, ¹⁷O recoupling is more favorable for RESPDOR experiments as the ¹¹B T_2' is significantly longer, and thus, more points at longer durations of recoupling could be obtained, which better define the dipolar dephasing build-up curve. Comparison of the experimental RESPDOR curves to those of numerical simulations of a two-spin ¹¹B-¹⁷O spin system enabled an estimate of the ¹⁷O isotopic abundance. Following this method for boric acid, the ¹⁷O isotopic abundance was estimated to be 30-33%, which matches well with the 30% determined from solution NMR spectroscopy. We also recorded ¹¹B{¹⁷O} *J*-resolved curves of boric acid and BDBA with CT-selective ¹⁷O inversion pulses to determine the one-bond ¹¹B-¹⁷O J-coupling constants (~35 Hz) and estimate the ¹⁷O isotopic abundance.

2D ¹¹B{¹⁷O} D-HMQC spectra of boric acid and BDBA were recorded at either $B_0 = 14.1$ T or 18.8 T with REDOR recoupling applied to the ¹¹B or ¹⁷O spins. In general, ¹¹B REDOR recoupling performed better than ¹⁷O REDOR recoupling; however, ¹⁷O recoupling was favorable in some cases. The choice of the recoupling channel is likely sampleand site-dependent. The effect that the ¹¹B and ¹⁷O $C_{\rm Q}$ have on the overall D-HMQC efficiency was investigated with SIMPSON numerical simulations. At 10 kHz MAS, the D-HMQC efficiency was relatively robust to both the $^{11}\mathrm{B}$ and $^{17}\mathrm{O}$ $C_{\rm Q}$ (0.5-4.5 MHz for ¹¹B and 4.0-9.0 MHz for ¹⁷O). Interestingly, numerical simulations predicted that ${}^{11}B{}^{17}O{}$ D-HMQC experiments operate at similar efficiencies at a variety of MAS frequencies (10-50 kHz). Lastly, we demonstrated the utility of using REDOR recoupling to enable HETCOR NMR experiments between two quadrupolar spins within materials by recording 2D ¹¹B{¹⁷O} and ²³Na- ${^{17}O}$ D-HMQC spectra of an ${^{17}O}$ -enriched sodium borate glass. The 2D ${^{11}B}{^{17}O}$ D-HMQC spectrum revealed unexpected heteronuclear correlations between ^[4]B and ¹⁷O NMR signals previously assigned to non-bridging O atoms. However, a 2D ¹¹B dipolar DQ-SQ homonuclear correlation NMR spectrum revealed strong [4]B-[4]B homonuclear correlations, illustrating that ^[4]B units linked via bridging O atoms must be common in this glass. Furthermore, periodic

plane-wave DFT calculations of boron oxide-based systems suggested that ${}^{[4]}B-O-{}^{[4]}B$ bridging O atoms generally exhibit a lower ${}^{17}O$ isotropic chemical shift as compared to ${}^{[3]}B-O-{}^{[3]}B$ or ${}^{[3]}B-O-{}^{[4]}B$ bridging O atoms. The heteronuclear correlations between ${}^{[4]}B$ and lower-frequency ${}^{17}O$ NMR signals were assigned to ${}^{[4]}B-O-{}^{[4]}B$ units. The ${}^{17}O$ NMR signals for ${}^{[3]}B-O^{-}$ and ${}^{[4]}B-O-{}^{[4]}B$ started to become slightly resolved at 35.2 T.

We anticipate that REDOR recoupling is generally applicable to an assortment of quadrupolar spins. The RESPDOR and D-HMQC experiments demonstrated here should be useful to study a variety of quadrupolar spin systems and materials.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jpcc.2c02737.

NMR experimental parameters, additional NMR spectra, and discussion of ¹⁷O isotopic enrichment determination (PDF)

AUTHOR INFORMATION

Corresponding Author

Aaron J. Rossini – Department of Chemistry, Iowa State University, Ames, Iowa 50011, United States; US Department of Energy, Ames Laboratory, Ames, Iowa 50011, United States; orcid.org/0000-0002-1679-9203; Phone: 515-294-8952; Email: arossini@iastate.edu

Authors

Rick W. Dorn – Department of Chemistry, Iowa State University, Ames, Iowa 50011, United States; US Department of Energy, Ames Laboratory, Ames, Iowa 50011, United States

- Alexander L. Paterson US Department of Energy, Ames Laboratory, Ames, Iowa 50011, United States; Present Address: National Magnetic Resonance Facility at Madison, University of Wisconsin–Madison, Madison, WI, USA 53706; • orcid.org/0000-0002-3069-5910
- Ivan Hung National High Magnetic Field Laboratory (NHMFL), Tallahassee, Florida 32310, United States; orcid.org/0000-0001-8916-739X
- Peter L. Gor'kov National High Magnetic Field Laboratory (NHMFL), Tallahassee, Florida 32310, United States
- Austin J. Thompson Department of Chemistry, Iowa State University, Ames, Iowa 50011, United States
- Aaron D. Sadow Department of Chemistry, Iowa State University, Ames, Iowa 50011, United States
- Zhehong Gan National High Magnetic Field Laboratory (NHMFL), Tallahassee, Florida 32310, United States;
 orcid.org/0000-0002-9855-5113

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.jpcc.2c02737

Notes

The authors declare no competing financial interest. Raw NMR data, SIMPSON numerical simulation files, MATLAB codes, and CASTEP DFT-calculated .cif and .magres files are available for download at DOI: 10.5281/zenodo.6376775

ACKNOWLEDGMENTS

This paper is dedicated to the memory of Professor Jacob Schaefer who invented REDOR solid-state NMR experiments. This work was primarily supported by the National Science Foundation under grant no. CBET-1916809. Borate glass synthesis (A.L.P.) was supported by the U.S. Department of Energy (DOE), Office of Science, Basic Energy Sciences, Materials Science and Engineering Division. The Ames Laboratory is operated for the U.S. DOE by Iowa State University under contract DE-AC02-07CH11358. Synthesis of ¹⁷O-labeled molecules (A.T. and A.D.S.) was supported by the National Science Foundation under grant no. CHE-1900393. A.J.R. acknowledges additional support from the Alfred P. Sloan Foundation through a Sloan research fellowship. This study made use of the National High Magnetic Field Laboratory to perform solid-state NMR spectroscopy experiments. The National High Magnetic Field Laboratory is supported by the National Science Foundation through NSF/ DMR-1644779 and the State of Florida. Development of the 36 T Series-Connected Hybrid (SCH) magnet and NMR instrumentation was supported by the NSF (DMR-1039938 and DMR-0603042) and NIH P41 GM122698. We thank Dr. Pierre Florian for helpful discussions and suggestions.

REFERENCES

(1) Hing, A. W.; Vega, S.; Schaefer, J. Transferred-Echo Double-Resonance NMR. J. Magn. Reson. 1992, 96, 205–209.

(2) Fyfe, C. A.; Mueller, K. T.; Grondey, H.; Wong-Moon, K. C. Dipolar Dephasing Between Quadrupolar and Spin-1/2 Nuclei. REDOR and TEDOR NMR Experiments on VPI-5. *Chem. Phys. Lett.* **1992**, *199*, 198–204.

(3) Grey, C. P.; Veeman, W. S. The Detection of Weak Heteronuclear Coupling Between Spin 1 and Spin 1/2 Nuclei in MAS NMR; $^{14}N/^{13}C/^{1}H$ Triple Resonance Experiments. *Chem. Phys. Lett.* **1992**, 192, 379–385.

(4) Hing, A. W.; Vega, S.; Schaefer, J. Measurement of Heteronuclear Dipolar Coupling by Transferred-Echo Double-Resonance NMR. J. Magn. Reson., Ser. A 1993, 103, 151–162.

(5) Grey, C. P.; Veeman, W. S.; Vega, A. J. Rotational echo ${}^{14}N/{}^{13}C/{}^{1}H$ triple resonance solid-state nuclear magnetic resonance: A probe of ${}^{13}C-{}^{14}N$ internuclear distances. J. Chem. Phys. 1993, 98, 7711–7724.

(6) Trebosc, J.; Hu, B.; Amoureux, J. P.; Gan, Z. Through-Space R3-HETCOR Experiments Between Spin-1/2 and Half-Integer Quadrupolar Nuclei in Solid-State NMR. *J. Magn. Reson.* **2007**, *186*, 220– 227.

(7) Cavadini, S.; Abraham, A.; Bodenhausen, G. Proton-Detected Nitrogen-14 NMR by Recoupling of Heteronuclear Dipolar Interactions using Symmetry-Based Sequences. *Chem. Phys. Lett.* **2007**, *445*, 1–5.

(8) Hu, B.; Trébosc, J.; Amoureux, J. P. Comparison of Several Hetero-Nuclear Dipolar Recoupling NMR Methods to be used in MAS HMQC/HSQC. J. Magn. Reson. 2008, 192, 112–122.

(9) Lafon, O.; Wang, Q.; Hu, B.; Vasconcelos, F.; Trébosc, J.; Cristol, S.; Deng, F.; Amoureux, J.-P. Indirect Detection via Spin-1/2 Nuclei in Solid State NMR Spectroscopy: Application to the Observation of Proximities between Protons and Quadrupolar Nuclei. *J. Phys. Chem. A* **2009**, *113*, 12864–12878.

(10) Martineau, C.; Bouchevreau, B.; Taulelle, F.; Trébosc, J.; Lafon, O.; Paul Amoureux, J. High-Resolution Through-Space Correlations Between Spin-1/2 and Half-Integer Quadrupolar Nuclei using the MQ-D-R-INEPT NMR Experiment. *Phys. Chem. Chem. Phys.* 2012, 14, 7112–7119.

(11) Haimovich, A.; Goldbourt, A. Characterization of Lithium Coordination Sites with Magic-Angle Spinning NMR. *J. Magn. Reson.* **2015**, *254*, 131–138.

(12) Venkatesh, A.; Hanrahan, M. P.; Rossini, A. J. Proton Detection of MAS Solid-State NMR Spectra of Half-Integer Quadrupolar Nuclei. *Solid State Nucl. Magn. Reson.* **2017**, *84*, 171–181.

(13) Szeto, K. C.; Merle, N.; Trébosc, J.; Taoufik, M.; Gauvin, R. M.; Pourpoint, F.; Delevoye, L. Caveat on the Actual Robustness of Heteronuclear NMR Methods for Probing the Surface of γ -Alumina and Related Catalysts. J. Phys. Chem. C 2019, 123, 12919–12927.

(14) Gullion, T.; Schaefer, J. Rotational-Echo Double-Resonance NMR. J. Magn. Reson. 1989, 81, 196–200.

(15) Gullion, T.; Schaefer, J. Detection of Weak Heteronuclear Dipolar Coupling by Rotational-Echo Double-Resonance Nuclear Magnetic Resonance. *Advances in Magnetic and Optical Resonance*; Warren, W. S., Ed.; Academic Press, 1989; Vol. 13, pp 57–83.

(16) Schmidt, A.; McKay, R. A.; Schaefer, J. Internuclear Distance Measurement Between Deuterium (I = 1) and a Spin-1/2 Nucleus in Rotating Solids. J. Magn. Reson. **1992**, 96, 644–650.

(17) Gullion, T. Introduction to Rotational-Echo, Double-Resonance NMR. Concepts Magn. Reson. 1998, 10, 277–289.

(18) Bertmer, M.; Eckert, H. Dephasing of Spin Echoes by Multiple Heteronuclear Dipolar Interactions in Rotational Echo Double Resonance NMR Experiments. *Solid State Nucl. Magn. Reson.* **1999**, *15*, 139–152.

(19) Lee, Y. K.; Kurur, N. D.; Helmle, M.; Johannessen, O. G.; Nielsen, N. C.; Levitt, M. H. Efficient Dipolar Recoupling in the NMR of Rotating Solids. A Sevenfold Symmetric Radiofrequency Pulse Sequence. *Chem. Phys. Lett.* **1995**, *242*, 304–309.

(20) Hohwy, M.; Jakobsen, H. J.; Edén, M.; Levitt, M. H.; Nielsen, N. C. Broadband Dipolar Recoupling in the Nuclear Magnetic Resonance of Rotating Solids: A Compensated C7 Pulse Sequence. J. Chem. Phys. **1998**, 108, 2686–2694.

(21) Edén, M.; Levitt, M. H. Pulse Sequence Symmetries in the Nuclear Magnetic Resonance of Spinning Solids: Application to Heteronuclear Decoupling. J. Chem. Phys. **1999**, 111, 1511–1519.

(22) Carravetta, M.; Edén, M.; Zhao, X.; Brinkmann, A.; Levitt, M. H. Symmetry Principles for the Design of Radiofrequency Pulse Sequences in the Nuclear Magnetic Resonance of Rotating Solids. *Chem. Phys. Lett.* **2000**, *321*, 205–215.

(23) Brinkmann, A.; Levitt, M. H. Symmetry Principles in the Nuclear Magnetic Resonance of Spinning Solids: Heteronuclear Recoupling by Generalized Hartmann–Hahn Sequences. J. Chem. Phys. 2001, 115, 357–384.

(24) Zhao, X.; Edén, M.; Levitt, M. H. Recoupling of Heteronuclear Dipolar Interactions in Solid-State NMR using Symmetry-Based Pulse Sequences. *Chem. Phys. Lett.* **2001**, *342*, 353–361.

(25) Levitt, M. H. Symmetry-Based Pulse Sequences in Magic-Angle Spinning Solid-State NMR. *eMagRes* 2007, DOI: 10.1002/9780470034590.emrstm0551.

(26) Gullion, T. Measurement of Dipolar Interactions Between Spin-1/2 and Quadrupolar Nuclei by Rotational-Echo, Adiabatic-Passage, Double-Resonance NMR. *Chem. Phys. Lett.* **1995**, *246*, 325– 330.

(27) Chopin, L.; Vega, S.; Gullion, T. A MAS NMR Method for Measuring $^{13}C-^{17}O$ Distances. J. Am. Chem. Soc. **1998**, 120, 4406–4409.

(28) Hughes, E.; Gullion, T.; Goldbourt, A.; Vega, S.; Vega, A. J. Internuclear Distance Determination of S=1, I=1/2 Spin Pairs Using REAPDOR NMR. *J. Magn. Reson.* **2002**, *156*, 230–241.

(29) Goldbourt, A.; Vega, S.; Gullion, T.; Vega, A. J. Interatomic Distance Measurement in Solid-State NMR between a Spin-1/2 and a Spin-5/2 Using a Universal REAPDOR Curve. J. Am. Chem. Soc. 2003, 125, 11194–11195.

(30) Gullion, T.; Vega, A. J. Measuring Heteronuclear Dipolar Couplings for I=1/2, S>1/2 Spin Pairs by REDOR and REAPDOR NMR. *Prog. Nucl. Magn. Reson. Spectrosc.* **2005**, 47, 123–136.

(31) Brinkmann, A.; Kentgens, A. P. M. Proton-Selective ¹⁷O-¹H Distance Measurements in Fast Magic-Angle-Spinning Solid-State

NMR Spectroscopy for the Determination of Hydrogen Bond Lengths. J. Am. Chem. Soc. 2006, 128, 14758–14759.

(32) Nimerovsky, E.; Goldbourt, A. Efficient Rotational Echo Double Resonance Recoupling of a Spin-1/2 and a Quadrupolar Spin at High Spinning Rates and Weak Irradiation Fields. *J. Magn. Reson.* **2010**, 206, 52–58.

(33) Nimerovsky, E.; Goldbourt, A. Distance Measurements Between Boron and Carbon at Natural Abundance using Magic Angle Spinning REAPDOR NMR and a Universal Curve. *Phys. Chem. Chem. Phys.* **2012**, *14*, 13437–13443.

(34) Carnahan, S. L.; Lampkin, B. J.; Naik, P.; Hanrahan, M. P.; Slowing, I. I.; VanVeller, B.; Wu, G.; Rossini, A. J. Probing O–H Bonding through Proton Detected ¹H–¹⁷O Double Resonance Solid-State NMR Spectroscopy. *J. Am. Chem. Soc.* **2019**, *141*, 441–450.

(35) Duong, N. T.; Rossi, F.; Makrinich, M.; Goldbourt, A.; Chierotti, M. R.; Gobetto, R.; Nishiyama, Y. Accurate ¹H-¹⁴N Distance Measurements by Phase-Modulated RESPDOR at Ultra-Fast MAS. *J. Magn. Reson.* **2019**, 308, 106559.

(36) Wijesekara, A. V.; Venkatesh, A.; Lampkin, B. J.; VanVeller, B.; Lubach, J. W.; Nagapudi, K.; Hung, I.; Gor'kov, P. L.; Gan, Z.; Rossini, A. J. Fast Acquisition of Proton-Detected HETCOR Solid-State NMR Spectra of Quadrupolar Nuclei and Rapid Measurement of NH Bond Lengths by Frequency Selective HMQC and RESPDOR Pulse Sequences. *Chem.—Eur. J.* **2020**, *26*, 7881–7888.

(37) Martins, V.; Xu, J.; Wang, X.; Chen, K.; Hung, I.; Gan, Z.; Gervais, C.; Bonhomme, C.; Jiang, S.; Zheng, A.; et al. Higher Magnetic Fields, Finer MOF Structural Information: ¹⁷O Solid-State NMR at 35.2 T. J. Am. Chem. Soc. **2020**, 142, 14877–14889.

(38) Wang, W. D.; Lucier, B. E. G.; Terskikh, V. V.; Wang, W.; Huang, Y. Wobbling and Hopping: Studying Dynamics of CO_2 Adsorbed in Metal–Organic Frameworks via ¹⁷O Solid-State NMR. J. Phys. Chem. Lett. **2014**, *5*, 3360–3365.

(39) Sutrisno, A.; Huang, Y. Solid-Atate NMR: A Powerful Tool for Characterization of Metal–Organic Frameworks. *Solid State Nucl. Magn. Reson.* **2013**, 49–50, 1–11.

(40) Xu, J.; Blaakmeer, E. S. M.; Lipton, A. S.; McDonald, T. M.; Liu, Y. M.; Smit, B.; Long, J. R.; Kentgens, A. P. M.; Reimer, J. A. Uncovering the Local Magnesium Environment in the Metal– Organic Framework Mg2(dobpdc) Using ²⁵Mg NMR Spectroscopy. J. Phys. Chem. C **2017**, *121*, 19938–19945.

(41) Dorn, R. W.; Ryan, M. J.; Kim, T.-H.; Goh, T. W.; Venkatesh, A.; Heintz, P. M.; Zhou, L.; Huang, W.; Rossini, A. J. Identifying the Molecular Edge Termination of Exfoliated Hexagonal Boron Nitride Nanosheets with Solid-State NMR Spectroscopy and Plane-Wave DFT Calculations. *Chem. Mater.* **2020**, *32*, 3109–3121.

(42) Wang, M.; Wu, X.-P.; Zheng, S.; Zhao, L.; Li, L.; Shen, L.; Gao, Y.; Xue, N.; Guo, X.; Huang, W.; et al. Identification of Different Oxygen Species in Oxide Nanostructures with ¹⁷O Solid-State NMR Spectroscopy. *Sci. Adv.* **2015**, *1*, No. e1400133.

(43) Zhang, W.; Bao, X.; Guo, X.; Wang, X. A High-Resolution Solid-State NMR Study on Nano-Structured HZSM-5 Zeolite. *Catal. Lett.* **1999**, *60*, 89–94.

(44) Scolan, E.; Magnenet, C.; Massiot, D.; Sanchez, C. Surface and Bulk Characterisation of Titanium–Oxo Clusters and Nanosized Titania Particles Through ¹⁷O Solid State NMR. *J. Mater. Chem.* **1999**, *9*, 2467–2474.

(45) Etman, A. S.; Pell, A. J.; Svedlindh, P.; Hedin, N.; Zou, X.; Sun, J.; Bernin, D. Insights into the Exfoliation Process of $V_2O_5 \cdot nH_2O$ Nanosheet Formation Using Real-Time ⁵¹V NMR. *ACS Omega* **2019**, *4*, 10899–10905.

(46) van Wüllen, L.; Züchner, L.; Müller-Warmuth, W.; Eckert, H. $^{11}B\{^{27}Al\}$ and $^{27}Al\{^{11}B\}$ Double Resonance Experiments on a Glassy Sodium Aluminoborate. *Solid State Nucl. Magn. Reson.* **1996**, *6*, 203–212.

(47) Aguiar, P. M.; Michaelis, V. K.; McKinley, C. M.; Kroeker, S. Network Connectivity in Cesium Borosilicate Glasses: ¹⁷O Multiple-Quantum MAS and Double-Resonance NMR. *J. Non-Cryst. Solids* **2013**, 363, 50–56.

(48) Funke, L. M.; Eckert, H. Charge Compensation in Sodium Borophosphate Glasses Studied by ${}^{11}B{}^{23}Na$ and ${}^{31}P{}^{23}Na$ Rotational Echo Double Resonance Spectroscopy. J. Phys. Chem. C 2016, 120, 3196–3205.

(49) Paterson, A. L.; Hannon, A. C.; Werner-Zwanziger, U.; Zwanziger, J. W. Structural Differences between the Glass and Crystal Forms of the Transparent Ferroelectric Nanocomposite, LaBGeO₅, from Neutron Diffraction and NMR Spectroscopy. *J. Phys. Chem.* C **2018**, *122*, 20963–20980.

(50) Eckert, H. Structural Characterization of Noncrystalline Solids and Glasses using Solid State NMR. *Prog. Nucl. Magn. Reson. Spectrosc.* **1992**, *24*, 159–293.

(51) Wang, Q.; Li, W.; Hung, I.; Mentink-Vigier, F.; Wang, X.; Qi, G.; Wang, X.; Gan, Z.; Xu, J.; Deng, F. Mapping the Oxygen Structure of γ -Al₂O₃ by High-Field Solid-State NMR Spectroscopy. *Nat. Commun.* **2020**, *11*, 3620.

(52) Ashbrook, S. E.; Smith, M. E. Solid State ¹⁷O NMR—An Introduction to the Background Principles and Applications to Inorganic Materials. *Chem. Soc. Rev.* **2006**, *35*, 718–735.

(53) Xu, J.; Wang, Q.; Deng, F. Metal Active Sites and Their Catalytic Functions in Zeolites: Insights from Solid-State NMR Spectroscopy. *Acc. Chem. Res.* **2019**, *52*, 2179–2189.

(54) Love, A. M.; Cendejas, M. C.; Thomas, B.; McDermott, W. P.; Uchupalanun, P.; Kruszynski, C.; Burt, S. P.; Agbi, T.; Rossini, A. J.; Hermans, I. Synthesis and Characterization of Silica-Supported Boron Oxide Catalysts for the Oxidative Dehydrogenation of Propane. *J. Phys. Chem. C* **2019**, *123*, 27000–27011.

(55) Chen, K.; Horstmeier, S.; Nguyen, V. T.; Wang, B.; Crossley, S. P.; Pham, T.; Gan, Z.; Hung, I.; White, J. L. Structure and Catalytic Characterization of a Second Framework Al(IV) Site in Zeolite Catalysts Revealed by NMR at 35.2 T. *J. Am. Chem. Soc.* **2020**, *142*, 7514–7523.

(56) Dorn, R. W.; Cendejas, M. C.; Chen, K.; Hung, I.; Altvater, N. R.; McDermott, W. P.; Gan, Z.; Hermans, I.; Rossini, A. J. Structure Determination of Boron-Based Oxidative Dehydrogenation Heterogeneous Catalysts With Ultrahigh Field 35.2 T ¹¹B Solid-State NMR Spectroscopy. *ACS Catal.* **2020**, *10*, 13852–13866.

(57) Hirsh, D. A.; Rossini, A. J.; Emsley, L.; Schurko, R. W. ³⁵Cl Dynamic Nuclear Polarization Solid-State NMR of Active Pharmaceutical Ingredients. *Phys. Chem. Chem. Phys.* **2016**, *18*, 25893–25904.

(58) Hong, Y.-I.; Manjunatha Reddy, G. N.; Nishiyama, Y. Selective Detection of Active Pharmaceutical Ingredients in Tablet Formulations using Solid-State NMR Spectroscopy. *Solid State Nucl. Magn. Reson.* **2020**, *106*, 101651.

(59) Pandey, M. K.; Kato, H.; Ishii, Y.; Nishiyama, Y. Two-Dimensional Proton-Detected ³⁵Cl/¹H Correlation Solid-State NMR Experiment Under Fast Magic Angle Sample Spinning: Application to Pharmaceutical Compounds. *Phys. Chem. Chem. Phys.* **2016**, *18*, 6209–6216.

(60) Hamaed, H.; Pawlowski, J. M.; Cooper, B. F. T.; Fu, R.; Eichhorn, S. H.; Schurko, R. W. Application of Solid-State ³⁵Cl NMR to the Structural Characterization of Hydrochloride Pharmaceuticals and their Polymorphs. *J. Am. Chem. Soc.* **2008**, *130*, 11056–11065.

(61) Veinberg, S. L.; Johnston, K. E.; Jaroszewicz, M. J.; Kispal, B. M.; Mireault, C. R.; Kobayashi, T.; Pruski, M.; Schurko, R. W. Natural Abundance ¹⁴N and ¹⁵N Solid-State NMR of Pharmaceuticals and their Polymorphs. *Phys. Chem. Chem. Phys.* **2016**, *18*, 17713–17730.

(62) Tatton, A. S.; Pham, T. N.; Vogt, F. G.; Iuga, D.; Edwards, A. J.; Brown, S. P. Probing Hydrogen Bonding in Cocrystals and Amorphous Dispersions Using $^{14}N^{-1}H$ HMQC Solid-State NMR. *Mol. Pharmaceutics* **2013**, *10*, 999–1007.

(63) Vogt, F. G.; Yin, H.; Forcino, R. G.; Wu, L. ¹⁷O Solid-State NMR as a Sensitive Probe of Hydrogen Bonding in Crystalline and Amorphous Solid Forms of Diflunisal. *Mol. Pharmaceutics* **2013**, *10*, 3433–3446.

(64) Tatton, A. S.; Pham, T. N.; Vogt, F. G.; Iuga, D.; Edwards, A. J.; Brown, S. P. Probing Intermolecular Interactions and Nitrogen Protonation in Pharmaceuticals by Novel ¹⁵N-Edited and 2D ¹⁴N-¹H Solid-State NMR. *CrystEngComm* **2012**, *14*, 2654–2659. (65) Iuga, D.; Morais, C.; Gan, Z.; Neuville, D. R.; Cormier, L.; Massiot, D. NMR Heteronuclear Correlation between Quadrupolar Nuclei in Solids. *J. Am. Chem. Soc.* **2005**, *127*, 11540–11541.

(66) Lee, S. K.; Deschamps, M.; Hiet, J.; Massiot, D.; Park, S. Y. Connectivity and Proximity between Quadrupolar Nuclides in Oxide Glasses: Insights from through-Bond and through-Space Correlations in Solid-State NMR. *J. Phys. Chem. B* **2009**, *113*, 5162–5167.

(67) Chen, C.-H.; Gaillard, E.; Mentink-Vigier, F.; Chen, K.; Gan, Z.; Gaveau, P.; Rebière, B.; Berthelot, R.; Florian, P.; Bonhomme, C.; et al. Direct ¹⁷O Isotopic Labeling of Oxides Using Mechanochemistry. *Inorg. Chem.* **2020**, *59*, 13050–13066.

(68) Sukenaga, S.; Florian, P.; Kanehashi, K.; Shibata, H.; Saito, N.; Nakashima, K.; Massiot, D. Oxygen Speciation in Multicomponent Silicate Glasses Using Through Bond Double Resonance NMR Spectroscopy. J. Phys. Chem. Lett. **201**7, *8*, 2274–2279.

(69) Chan, J. C. C.; Bertmer, M.; Eckert, H. Double-Quantum Cross-Polarization Between Half-Integer Quadrupolar Spin Systems: ${}^{11}B\leftrightarrow{}^{23}Na$ and ${}^{11}B\leftrightarrow{}^{27}Al$. *Chem. Phys. Lett.* **1998**, *292*, 154–160.

(70) Chan, J. C. C. High-Resolution Heteronuclear Correlation between Quadrupolar Nuclei. J. Magn. Reson. 1999, 140, 487–490.

(71) Chan, J. C. C.; Bertmer, M.; Eckert, H. Site Connectivities in Amorphous Materials Studied by Double-Resonance NMR of Quadrupolar Nuclei: High-Resolution ¹¹B \leftrightarrow ²⁷Al Spectroscopy of Aluminoborate Glasses. J. Am. Chem. Soc. **1999**, 121, 5238–5248.

(72) Lu, X.; Tankamony, A. S. L.; Trébosc, J.; Lafon, O.; Amoureux, J.-P. Probing Proximities Between Different Quadrupolar Isotopes using Multi-Pulse Cross-Polarization. *J. Magn. Reson.* **2013**, *228*, 148–158.

(73) Vega, A. J. CPMAS of Quadrupolar S = 3/2 Nuclei. Solid State Nucl. Magn. Reson. 1992, 1, 17–32.

(74) Vega, A. J. MAS NMR Spin Locking of Half-Integer Quadrupolar Nuclei. J. Magn. Reson. **1992**, 96, 50–68.

(75) van Wüllen, L. Double Resonance Experiments in a Single Resonance Probe: Detecting ${}^{23}\text{Na}-{}^{51}\text{V}$ Dipolar Interactions in Sodium Vanadates. *Solid State Nucl. Magn. Reson.* **1998**, *10*, 235–240. (76) Janssen, M.; Eckert, H. ${}^{11}\text{B}{}^{23}\text{Na}{}$ Rotational Ccho Double Resonance NMR: A New Approach for Studying the Spatial Cation Distribution in Sodium Borate Glasses. *Solid State Ionics* **2000**, *136–137*, 1007–1014.

(77) Yang, J.; Zheng, A.; Zhang, M.; Luo, Q.; Yue, Y.; Ye, C.; Lu, X.; Deng, F. Brönsted and Lewis Acidity of the BF_3/γ -Al₂O₃ Alkylation Catalyst as Revealed by Solid-State NMR Spectroscopy and DFT Quantum Chemical Calculations. *J. Phys. Chem. B* **2005**, *109*, 13124–13131.

(78) Hansen, M. R.; Jakobsen, H. J.; Skibsted, J. Structural Environments for Boron and Aluminum in Alumina–Boria Catalysts and Their Precursors from ¹¹B and ²⁷Al Single- and Double-Resonance MAS NMR Experiments. *J. Phys. Chem. C* **2008**, *112*, 7210–7222.

(79) Xin, S.; Wang, Q.; Xu, J.; Feng, N.; Li, W.; Deng, F. Heteronuclear Correlation Experiments of ²³Na-²⁷Al in Rotating Solids. *Solid State Nucl. Magn. Reson.* **2017**, *84*, 103–110.

(80) Zheng, M.; Xin, S.; Wang, Q.; Trébosc, J.; Xu, J.; Qi, G.; Feng, N.; Lafon, O.; Deng, F. Through-Space ${}^{11}B-{}^{27}Al$ Correlation: Influence of the Recoupling Channel. *Magn. Reson. Chem.* **2021**, *59*, 1062–1076.

(81) Harris, R. K.; Becker, E. D.; Cabral de Menezes, S. M.; Goodfellow, R.; Granger, P. NMR Nomenclature: Nuclear Spin Properties and Conventions for Chemical Shifts: IUPAC Recommendations 2001. *Solid State Nucl. Magn. Reson.* 2002, *22*, 458–483. (82) van Meerten, S. G. J.; Franssen, W. M. J.; Kentgens, A. P. M. ssNake: A Cross-Platform Open-Source NMR Data Processing and Fitting Application. *J. Magn. Reson.* 2019, *301*, 56–66.

(83) Frydman, L.; Harwood, J. S. Isotropic Spectra of Half-Integer Quadrupolar Spins from Bidimensional Magic-Angle Spinning NMR. J. Am. Chem. Soc. **1995**, 117, 5367–5368.

(84) Medek, A.; Harwood, J. S.; Frydman, L. Multiple-Quantum Magic-Angle Spinning NMR: A New Method for the Study of Quadrupolar Nuclei in Solids. J. Am. Chem. Soc. 1995, 117, 12779–12787.

(85) Brown, S. P.; Wimperis, S. Two-Dimensional Multiple-Quantum MAS NMR of Quadrupolar Nuclei. Acquisition of the Whole Echo. J. Magn. Reson. **1997**, *124*, 279–285.

(86) Hwang, S.-J.; Fernandez, C.; Amoureux, J. P.; Cho, J.; Martin, S. W.; Pruski, M. Quantitative Study of the Short Range Order in B_2O_3 and B_2S_3 by MAS and Two-Dimensional Triple-Quantum MAS ¹¹B NMR. *Solid State Nucl. Magn. Reson.* **1997**, *8*, 109–121.

(87) Yao, Z.; Kwak, H.-T.; Sakellariou, D.; Emsley, L.; Grandinetti, P. J. Sensitivity Enhancement of the Central Transition NMR Signal of Quadrupolar Nuclei under Magic-Angle Spinning. *Chem. Phys. Lett.* **2000**, 327, 85–90.

(88) Kwak, H.-T.; Prasad, S.; Clark, T.; Grandinetti, P. J. Enhancing Sensitivity of Quadrupolar Nuclei in Solid-State NMR with Multiple Rotor Assisted Population Transfers. *Solid State Nucl. Magn. Reson.* **2003**, *24*, 71–77.

(89) Fung, B. M.; Khitrin, A. K.; Ermolaev, K. An Improved Broadband Decoupling Sequence for Liquid Crystals and Solids. J. Magn. Reson. 2000, 142, 97–101.

(90) Venkatesh, A.; Perras, F. A.; Rossini, A. J. Proton-Detected Solid-State NMR Spectroscopy of Spin-1/2 Nuclei with Large Chemical Shift Anisotropy. J. Magn. Reson. **2021**, 327, 106983.

(91) Wang, Q.; Hu, B.; Lafon, Ö.; Trébosc, J.; Deng, F.; Amoureux, J. P. Double-Quantum Homonuclear NMR Correlation Spectroscopy of Quadrupolar Nuclei Subjected to Magic-Angle Spinning and High Magnetic Field. J. Magn. Reson. 2009, 200, 251–260.

(92) Mali, G.; Fink, G.; Taulelle, F. Double-Quantum Homonuclear Correlation Magic Angle Sample Spinning Nuclear Magnetic Resonance Spectroscopy of Dipolar-Coupled Quadrupolar Nuclei. J. Chem. Phys. **2004**, 120, 2835–2845.

(93) Gan, Z.; Hung, I.; Wang, X.; Paulino, J.; Wu, G.; Litvak, I. M.; Gor'kov, P. L.; Brey, W. W.; Lendi, P.; Schiano, J. L.; et al. NMR Spectroscopy up to 35.2T using a Series-Connected Hybrid Magnet. J. Magn. Reson. **2017**, 284, 125–136.

(94) Bak, M.; Rasmussen, J. T.; Nielsen, N. C. SIMPSON A General Simulation Program for Solid-State NMR Spectroscopy. *J. Magn. Reson.* **2000**, *147*, 296–330.

(95) Tošner, Z.; Vosegaard, T.; Kehlet, C.; Khaneja, N.; Glaser, S. J.; Nielsen, N. C. Optimal Control in NMR Spectroscopy: Numerical Implementation in SIMPSON. *J. Magn. Reson.* **2009**, *197*, 120–134.

(96) Tošner, Z.; Andersen, R.; Stevensson, B.; Edén, M.; Nielsen, N. C.; Vosegaard, T. Computer-Intensive Simulation of Solid-State NMR Experiments using SIMPSON. *J. Magn. Reson.* **2014**, *246*, 79–93.

(97) Pickard, C. J.; Mauri, F. All-Electron Magnetic Response with Pseudopotentials: NMR Chemical Shifts. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2001**, *63*, 245101.

(98) Clark, S. J.; Segall, M. D.; Pickard, C. J.; Hasnip, P. J.; Probert, M. I. J.; Refson, K.; Payne, M. C. First Principles Methods using CASTEP. Z. Kristallogr.—Cryst. Mater. **2005**, 220, 567–570.

(99) Pfrommer, B. G.; Côté, M.; Louie, S. G.; Cohen, M. L. Relaxation of Crystals with the Quasi-Newton Method. *J. Comput. Phys.* **1997**, *131*, 233–240.

(100) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. *Phys. Rev. Lett.* **1996**, *77*, 3865–3868.

(101) Tkatchenko, A.; Scheffler, M. Accurate Molecular Van Der Waals Interactions from Ground-State Electron Density and Free-Atom Reference Data. *Phys. Rev. Lett.* **2009**, *102*, 073005.

(102) Vanderbilt, D. Soft Self-Consistent Pseudopotentials in a Generalized Eigenvalue Formalism. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1990**, *41*, 7892–7895.

(103) Yates, J. R.; Pickard, C. J.; Mauri, F. Calculation of NMR Chemical Shifts for Extended Systems using Ultrasoft Pseudopotentials. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2007**, *76*, 024401.

(104) Green, T. F. G.; Yates, J. R. Relativistic Nuclear Magnetic Resonance J-coupling with Ultrasoft Pseudopotentials and the Zeroth-Order Regular Approximation. J. Chem. Phys. **2014**, 140, 234106.

(105) Garbow, J. R.; Gullion, T. The Importance of Precise Timing in Pulsed, Rotor-Synchronous MAS NMR. *Chem. Phys. Lett.* **1992**, *192*, 71–76.

(106) Shuvalov, R. R.; Burns, P. C. A New Polytype of Orthoboric Acid, H₃BO₃-3T. Acta Crystallogr., Sect. C: Cryst. Struct. Commun. **2003**, 59, i47–i49.

(107) Rodríguez-Cuamatzi, P.; Vargas-Díaz, G.; Maris, T.; Wuest, J. D.; Höpfl, H. 1,4-Phenylenediboronic Acid. *Acta Crystallogr., Sect. E: Struct. Rep. Online* **2004**, *60*, o1316–o1318.

(108) Wrackmeyer, B.; Tok, O. L. Indirect Nuclear Spin-Spin Coupling Constants ¹J(¹⁷O,¹¹B). First Observation and Calculation using Density Functional Theory (DFT). *Z. Naturforsch., B: J. Chem. Sci.* **2006**, *61*, 949–955.

(109) Wang, Q.; Trébosc, J.; Li, Y.; Xu, J.; Hu, B.; Feng, N.; Chen, Q.; Lafon, O.; Amoureux, J.-P.; Deng, F. Signal enhancement of J-HMQC experiments in solid-state NMR involving half-integer quadrupolar nuclei. *Chem. Commun.* **2013**, *49*, 6653–6655.

(110) Wang, Q.; Li, Y.; Trébosc, J.; Lafon, O.; Xu, J.; Hu, B.; Feng, N.; Chen, Q.; Amoureux, J.-P.; Deng, F. Population Transfer HMQC for Half-Integer Quadrupolar Nuclei. *J. Chem. Phys.* **2015**, *142*, 094201.

(111) Stebbins, J. F.; Zhao, P.; Kroeker, S. Non-Bridging Oxygens in Borate Glasses: Characterization by ¹¹B and ¹⁷O MAS and 3QMAS NMR. *Solid State Nucl. Magn. Reson.* **2000**, *16*, 9–19.

(112) Kriz, H. M.; Bray, P. J. The ¹¹B Quadrupole Interaction and Nonbridging Oxygens in Crystalline Borates. J. Magn. Reson. 1971, 4, 76–84.

(113) Kriz, H. M.; Bishop, S. G.; Bray, P. J. ¹¹B NMR in Polycrystalline Calcium Metaborate. *J. Chem. Phys.* **1968**, *49*, 557–561.

(114) Kroeker, S.; Stebbins, J. F. Three-Coordinated Boron-11 Chemical Shifts in Borates. *Inorg. Chem.* **2001**, *40*, 6239–6246.

(115) Aguiar, P. M.; Kroeker, S. Boron Speciation and Non-Bridging Oxygens in High-Alkali Borate Glasses. *J. Non-Cryst. Solids* **2007**, *353*, 1834–1839.

(116) Feller, S. A.; Dell, W. J.; Bray, P. J. ¹⁰B NMR studies of lithium borate glasses. *J. Non-Cryst. Solids* **1982**, *51*, 21–30.

(117) Yu, Y.; Stevensson, B.; Edén, M. Direct Experimental Evidence for Abundant BO_4-BO_4 Motifs in Borosilicate Glasses From Double-Quantum ¹¹B NMR Spectroscopy. *J. Phys. Chem. Lett.* **2018**, *9*, 6372–6376.

Recommended by ACS

Subsecond Three-Dimensional Nitrogen-15 Magnetic Resonance Imaging Facilitated by Parahydrogen-Based Hyperpolarization

Alexandra I. Trepakova, Igor V. Koptyug, *et al.* OCTOBER 27, 2022 THE JOURNAL OF PHYSICAL CHEMISTRY LETTERS

READ 🗹

Ultrahigh-Resolution Homo- and Heterodecoupled ¹H and TOCSY NMR Experiments

István Timári, Katalin E. Kövér, et al. NOVEMBER 15, 2022 ACS OMEGA

READ 🗹

Through-Space Multinuclear Magnetic Resonance Signal Enhancement Induced by Parahydrogen and Radiofrequency Amplification by Stimulated Emission of...

Oleg G. Salnikov, Igor V. Koptyug, *et al.* OCTOBER 20, 2022 ANALYTICAL CHEMISTRY

READ 🗹

RFAD

Sensitivity Enhancement by Progressive Saturation of the Proton Reservoir: A Solid-State NMR Analogue of Chemical Exchange Saturation Transfer

Michael J. Jaroszewicz, Lucio Frydman, et al. NOVEMBER 18, 2021 JOURNAL OF THE AMERICAN CHEMICAL SOCIETY

Get More Suggestions >